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Biotransformation of Neoruscogenin by the Endophytic Fungus Alternaria eureka

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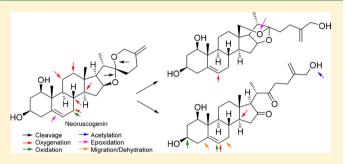
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Supporting Information

ABSTRACT: Biotransformation of neoruscogenin (NR, 1, spirosta-5,25(27)-diene- 1β , 3β -diol), the major bioactive sapogenin of *Ruscus* preparations, was carried out with the endophytic fungus *Alternaria eureka*. Fourteen new biotransformation products (2–15) were isolated, and their structures were elucidated by NMR and HRESIMS data analyses. *A. eureka* affected mainly oxygenation, oxidation, and epoxidation reactions on the B and C rings of the sapogenin to afford compounds 8–15. In addition to these, cleavage of the spiroketal system as in compounds 2–7 and subsequent transformations provided unusual metabolites. This is the first



study reporting conversion of the spirostanol skeleton to cholestane-type metabolites 2-5. Additionally, the cleavage of the C-22/C-26 oxygen bridge yielding a furostanol-type steroidal framework and subsequent formation of the epoxy bridge between C-18 and C-22 in 7 was encountered for the first time in steroid chemistry.

The natural product drug discovery process involves isolation of new metabolites from natural sources, investigation of their biological activities, and semisynthesis of more active analogues. Compared to chemical methods microbial biotransformation is advantageous for preparing new oxygenated steroid derivatives due to the stereo- and regioselective reaction capability and its operation under mild conditions.¹⁻⁴ In addition to overcoming the problems confronted in chemical synthesis, this process can be of assistance to establish biosynthetic pathways, increase/decrease the bioactivity/toxicity profiles of the lead molecules, or support structure–activity relationship studies. Moreover, it is a useful method to predict drug metabolism due to the similarity of the enzyme systems of fungi and mammalians.^{5,6}

Endophytic organisms are bacterial or fungal microorganisms that colonize internal tissues of living plants without causing any disease symptoms, and they constitute one of the most exciting groups of microorganisms. They have attracted a great deal of attention due to their ability to produce novel bioactive secondary metabolites for many years.^{7–9} While producing interesting chemistry, they are also able to interact with their hosts via special enzymes and chemicals, which make them useful sources as biocatalysts.^{10–15}

Neoruscogenin (NR, 1) is one of the major spirostanol steroids of *Ruscus aculeatus* (Butcher's broom, Rusci rhizome, Asparagaceae). The powdered roots, extracts, and the mixture of the major bioactive steroidal sapogenins ruscogenins [mixture of two major sapogenins, viz., neoruscogenin (1) and ruscogenin] are components of various pharmaceutical preparations to treat chronic venous insufficiency, varicose veins, hemorrhoids, and orthostatic hypotension due to their pharmacological activities.^{16–24} The European Pharmacopoeia prescribes "not less than 1.0% of total sapogenins expressed as ruscogenins (neoruscogenin and ruscogenin mixture)", and the European Scientific Cooperative on Phytotherapy recommends oral use of 7–11 mg/day.^{25,26}

Despite several reported semisynthesis and biotransformation studies of steroids, studies performed on ruscogenin have so far been limited, 2^{2-30} and there has been neither an in vitro nor in vivo report on NR revealing its metabolic fate. In order to shed light into the metabolism of neoruscogenin (1), estimate its likely mammalian metabolites, and investigate the potential use of endophytes as biocatalyst, we herein report biotransformation of NR by the endophytic fungus Alternaria eureka. As a result, 14 biotransformation products (2-15) were purified from the biotransformation broth by extensive chromatographic procedures. The structures of the metabolites were elucidated by NMR and HRESIMS data analyses. A. eureka affected mainly the anticipated oxygenation, oxidation, and epoxidation reactions to afford compounds 8-15, whereas cleavage of the spiroketal system resulted in exceptional transformation products 2-7.

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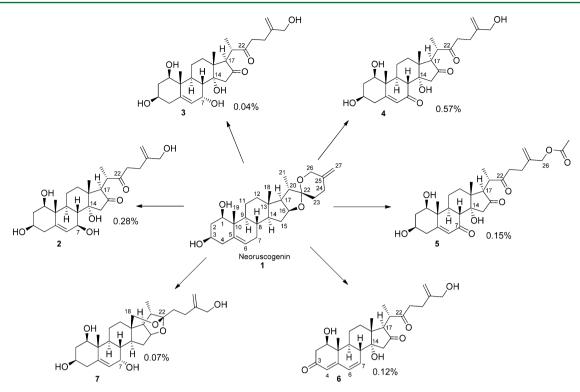
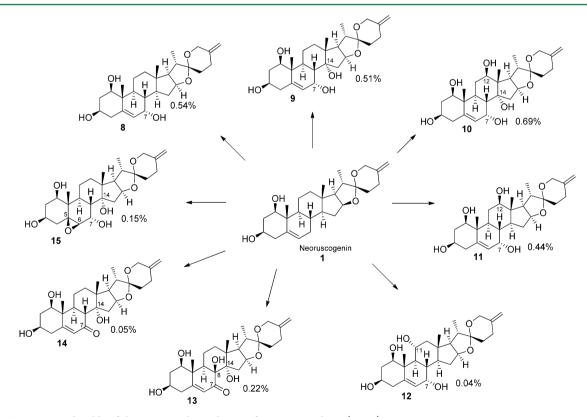


Figure 1. Structures and yields of the cholestane- (2-6) and furostanol-type (7) biotransformation products.



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Figure 2. Structures and yields of the spirostanol-type biotransformation products (8–15).

RESULTS AND DISCUSSION

The literature survey and preliminary screening studies have prompted the use of the endophytic fungus *A. eureka* as biocatalyst to transform the steroidal sapogenin neoruscogenin (1). After 10 days of incubation, 14 metabolites (2-15) (Figures 1 and 2) were isolated from the broth, and the structures were elucidated by means of 1D and 2D NMR and HRMS experiments. The 1 H and 13 C NMR spectroscopic data of the compounds are collated in Tables 1–3.

The sodium adduct ion at m/z 499.2653 [M + Na]⁺ (calcd 499.2671 for C₂₇H₄₀O₇Na) in the HRESIMS spectrum of 2

Article

Table 1. ¹H and ¹³C NMR Data of Compounds (2–6) (500 MHz, in Pyridine-d₅)

		2		3		4	5		6	
position	$\delta_{ m C}$	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{ m C}$	$\delta_{ m H}~(J~{ m in~Hz})$	δ_{C}	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	$\delta_{ m C}$	$\delta_{ m H}~(J~{ m in~Hz})$	δ_{C}	$\delta_{ m H}~(J~{ m in~Hz})$
1	77.2	3.84 d (11.5)	77.5	3.82 m	76.2	3.83 m 76.2 3.87 d (11.0)		72.5	4.27 ddd (5.5, 6.0, 6.0)	
2	43.7	2.27 m, 2.65 m	42.8	2.24 ddd (11.5, 12.0, 12.0), 2.53 m	42.9	2.27 dd (11.5, 12.0), 2.62 m			44.9	2.99 m, 3.05 m
3	67.7	3.95 m	67.3	3.84 m	66.1	3.93 m	66.0	3.91 m	197.7	
4	43.6	2.70 m	43.7	2.74-2.78 m	43.6	2.76 m	43.6	2.75 m	124.3	5.93 s
5	140.2		143.9		166.0		166.2		162.8	
6	131.0	5.97 s	126.1	5.93 d (4.5)	128.2	6.04 s	128.2	6.03 s	130.1	6.30 dd (1.0, 9.5)
7	66.7	4.96 bs	66.1	4.32 s	200.1		200.5		140.1	6.25 dd (1.0, 9.5)
8	43.6	2.27 m	37.9	1.95 m	48.3	2.87 m	48.3	2.86 m	41.3	2.89 m
9	42.9	2.55 ddd (5.0, 5.0, 11.5)	38.2	2.77 m	44.8	2.94 m	2.94 m 44.8 2.94 m		44.7	2.61 m
10	43.2		44.1		44.8		44.8		42.4	
11	22.9	1.95 m, 2.99 m	22.7	1.93 m, 2.92 m	23.0	1.99 ddd (13.0, 12.0, 12.5), 2.92 m	23.1	1.99 m, 2.92 m	23.3	1.99 m, 2.55 m
12	31.4	1.70 d (12.0), 2.67 m	31.4	1.60 m, 2.72 m	30.9	1.68 d (12.5), 2.62 m	30.8	1.67 m, 2.61 m	31.5	1.70 m, 2.53 m
13	46.1		46.0		46.6		45.6		46.0	
14	78.4		79.7		77.1		77.1		78.6	
15	49.8	3.20 d (18.5), 3.05 d (18.5)	47.5	2.49 d (17.5), 2.64 d (17.5)	49.2	2.75 m, 3.59 d (19.2)	49.2	3.57 m	46.3	2.60 m, 2.75 m
16	220.1		218.7		219.3		219.4		217.9	
17	60.2	3.92 d (11.0)	60.7	3.89 d (11.0)	59.5	3.87 d (11.0)	59.6	3.87 d (11.0)	60.6	3.94 d (13.0)
18	16.6	1.07 s	16.3	1.00 s	16.9	1.01 s	17.2	1.02 s	17.2	1.07 s
19	13.4	1.41 s	12.9	1.37 s	12.1	1.50 s	12.1 1.50 s		11.4	1.40 s
20	43.7	2.80 m	43.4	2.75 m	43.4	2.74 m	43.5	2.77 m	43.7	2.80 m
21	15.7	1.08 d (5.5)	15.6	1.05 d (7.0)	15.6	1.06 d (5.6)	15.3	1.08 d (6.5)	15.5	1.06 d (7.0)
22	212.9		212.6		212.8		212.8		212.5	
23	40.8	3.00 m, 3.13 dd (7.5, 8.5)	40.7	3.00 m, 3.15 m	40.8	2.95 m, 3.16 m	40.3	2.93 m, 3.09 m	40.8	3.00 m, 3.13 m
24	27.0	2.70 m	27.1	2.74 m	27.0	2.72 m	26.9	2.57 m	27.1	2.22 m (2H)
25	150.1		150.2		150.1		144.2		150.2	
26	65.2	4.39 s	65.2	4.40 s	65.2	4.39 s	66.9	4.69 s	65.3	4.41 s
27	108.3	5.07 s, 5.42 s	108.2	5.08 s, 5.43 s	108.3	5.43 s, 5.08 s	112.0	5.07 s, 5.13 s	107.9	5.09 s, 5.43 s
<u>С</u> ОС <u>Н</u> 3 СО <u>С</u> Н3							170.4 20.5	2.00 s		

indicated eight indices of hydrogen deficiency, a 48 amu increase over 1. The ¹³C NMR and DEPT135 spectra showed two carbonyl signals at $\delta_{\rm C}$ 212.9 and 220.1, two resonances of a disubstituted olefinic system ($\delta_{\rm C}$ 108.3, t; 150.1, s), and signals of a trisubstituted olefinic system ($\delta_{\rm C}$ 131.0, d; 140.2, s) that revealed a tetracyclic framework for 2, suggesting ring cleavage(s). Based on the absence of the characteristic spiroketal carbon resonance (C-22, ca. 110 ppm) and the characteristic H-16 signal, the low-field shift of the exocyclic H₂-27 resonances (ca. 0.5 ppm), and replacement of the characteristic AB-system signal of H₂-26 with an oxymethylene singlet, the aforementioned cleavage was evident. Thus, it was inferred that the spirostanol skeleton was transformed into a steroidal derivative with an acyclic side chain. The DEPT-135, ¹³C NMR, and HSQC spectra showed two additional low-field signals at $\delta_{\rm C}$ 78.4 (s) and 66.7 (d). The former carbon displayed a long-range HMBC cross-peak with an exchangeable proton resonating at $\delta_{\rm H}$ 6.48 (Figure 3), indicative of hydroxylation of a methine carbon of 1. Examination of the COSY and TOCSY spectra led to identification of five main spin systems (SS1-SS5) in 2. One of these spin systems, SS-2 $[H-6 \rightarrow H-7(O) \rightarrow H-8 \rightarrow H-9 \rightarrow H_2-11 \rightarrow H_2-12]$, located

one of the hydroxy groups at C-7. The deshielding of H-7 ($\delta_{\rm H}$ 4.96) due to the $\Delta^{5(6)}$ double bond was also noteworthy. The simplified H-17 spin system evident in the COSY spectrum and the 1.00 ppm deshielding of H-17 ($\delta_{\rm H}$ 3.92, d, J = 11.0 Hz) compared to 1 implied the presence of a carbonyl group at C-16. In the HMBC spectrum, the carbonyl resonating at $\delta_{\rm C}$ 220.1 (C-16) displayed correlations with H-17 and protons of a methylene group (H₂-15) resonating as an AB system ($\delta_{\rm H}$ 3.20 and 3.05, both d, J_{AB} = 18.5 Hz). The isolated C-15 methylene protons indicated functionalization at C-14. The HMBC correlation of the carbon at $\delta_{\rm C}$ 78.4 (s) and H₃-18 defined the location of the second hydroxy group at C-14. The spin systems and their interconnectivities secured by the 2D NMR spectra permitted establishment of the tetracyclic skeleton from C-1 to C-21. The aforementioned conclusions left the C-22 \rightarrow C-27 system to be identified. The HMBC correlation of the second carbonyl signal ($\delta_{\rm C}$ 212.9) with H₃-21 ($\delta_{\rm H}$ 1.08, d, J = 5.5 Hz) and H-20 ($\delta_{\rm H}$ 2.80) verified that C-22 was also oxygenated after cleavage of the spiroketal system to produce the C-22 carbonyl group. Inspection of the COSY spectrum revealed another spin system (SS-4) including two methylene groups, which was readily assigned to H₂-23 and H₂-24 based

Table 2. ¹H and ¹³C NMR Data of Compounds 7–11 (500 MHz, in Pyridine-d₅)

	7		8		9		10		11	
position	$\delta_{\rm C}$	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	δ_{C}	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{ m C}$	$\delta_{\rm H} \left(J \text{ in Hz} \right)$	$\delta_{\rm C}$	$\delta_{ m H}~(J~{ m in}~{ m Hz})$	$\delta_{\rm C}$	$\delta_{ m H}$ (J in Hz)
1	77.6	3.83 d (8.5)	77.5	3.82 dd (3.5, 11.5)	77.5	3.80 dd (4.0, 11.5)	77.5	3.84 m	77.4	3.83 m
2	43.0	2.23 m, 2.55 m	43.1	2.55 m, 2.23 ddd (12, 11.5, 11.5)	42.9	2.23 m, 2.51 m	42.8	2.24 m, 2.53 m	42.7	2.25 m, 2.55 m
3	67.4	3.89 m	67.4	3.89 m	67.3	3.83 m	67.3	3.84 m	67.4	3.89 m
4	43.5	2.70 m	43.6	2.70 m	43.6	2.75 m	43.4	2.70 m	43.5	2.72 m
5	142.9		143.1		143.7		143.7		143.0	
6	127.8	6.04 d (5.0)	128.0	6.06 d (5.0)	126.5	5.94 d (5.0)	126.5	5.95 d (5.0)	127.7	6.07 d (5.0)
7	64.3	4.07 s	64.7	4.09 bs	66.3	4.46 s	66.2	4.51 d (12.0)	64.7	4.14 bs
8	38.7	1.77 m	38.5	1.77 m	39.4	2.02 m	38.7	2.07 dd (3.5, 7.5)	37.6	1.77 m
9	43.1	2.14 m	43.6	2.70 m	38.3	2.76 m	38.7	2.87 ddd (5.5, 12.5, 13.0)	42.7	2.35 m
10	44.0		44.1	44.1	44.2		44.3		44.2	
11	24.0	1.41 m, 2.97 dd (3.5, 14.0)	23.9	1.83 m, 2.90 m	23.1	1.93 m, 2.92 m	33.6	2.14 m, 3.36 m	34.1	2.10 m, 3.34 m
12	37.4	1.31 m, 1.88 m	40.2	1.33 m, 1.77 m	32.4	1.42 m, 2.51 m	72.4	4.96 dd (5.0, 11.6)	79.2	3.80 m
13	45.2		39.9		44.5		49.7		45.5	
14	48.3	2.11 m	50.2	2.10 m	87.5		89.7		49.0	2.17 m
15	36.8	2.69 m, 1.83 m	32.1	1.61 m, 2.55 m	40.8	2.04 m, 2.62 m	41.2	2.20 m, 2.66 m	31.9	1.81 m, 2.65 m
16	80.5	4.64 m	81.5	4.61 ddd (7.5, 11.5, 12.5)	82.2	5.12 ddd (6.5, 7.0, 7.5)	82.4 5.22 ddd (6.0, 7.5, 8.0)		81.5	4.69 ddd (7.5, 8.0, 8.5)
17	56.6	1.82 m	63.0	1.88 m	59.5	2.80 m	59.4	3.18 dd (8.0, 7.0)	62.9	2.25 m
18	65.2	3.41 d (11.8), 4.07 d (11.9)	16.4	0.98 s	19.7	1.12 s	13.5	1.37 s	11.1	1.24 s
19	12.5	1.26 s	12.7	1.34 s	13.0	1.36 s	13.0	1.38 s	12.6	1.36 s
20	35.9	2.23 m	41.7	1.99 dd (6.5, 7.0)	41.7	2.10 m	42.6	2.31 m	43.0	2.25 m
21	14.5	0.85 d (6.5)	14.8	1.04 d (7.0)	15.1	1.09 d (7.0)	14.4	1.41 d (6.0)	14.1	1.36 d (6.0)
22	109.9		109.2		109.4		109.8		109.5	
23	27.8	2.47 m, 2.61 m	33.0	1.78 m	33.1	1.82 m	33.2	1.87 m	31.9	1.79 m, 2.65 m
24	32.8	2.03 m, 2.23 m	28.7	2.23 m, 2.70 m	28.8	2.25 m, 2.74 m	28.9	2.25 m, 2.77 m	29.8	2.24 m, 2.73 m
25	151.1		144.3		144.3		144.4		144.3	
26	65.2	4.43 s	64.3	4.00 d (12.0), 4.40 d (12.0)	64.7	4.00 d (9.5), 4.47 d (11.5)	64.7	4.05 d (12.0), 4.51 d (12.0)	64.3	4.00 d (12.0), 4.44 d (12.0)
27	107.8	5.04 s, 5.38 s	108.4	4.78 d (9.5)	108.5	4.87 d (9.5)	108.5	4.78 s, 4.81 s	180.4	4.79 s, 4.77 s

on their ${}^{2}J_{H-C}$ and ${}^{3}J_{H-C}$ correlations with C-22 ($\delta_{\rm C}$ 212.9), respectively. The long-range correlations of the terminal hydroxymethyl signal at $\delta_{\rm H}$ 4.39 (s) with the carbon resonances at $\delta_{\rm C}$ 150.1 (C-25), 108.3 (C-27), and 27.0 (C-24), and C-24, C-25 and C-26 to H₂-27 ($\delta_{\rm H}$ 5.07 and 5.42, both s), were of assistance to deduce the remaining part of the side chain as shown in Figure 3. The relative configuration was established via the 2D-ROESY data. The correlation of H-7 with α -oriented H-9 ($\delta_{\rm H}$ 2.55, ddd, J = 5.0, 5.0, 11.5 Hz) and the cross-peak between these two resonances and exchangeable proton signal at $\delta_{\rm H}$ 6.48 [C-14(OH)] confirmed the orientation of the OH groups as 7β and 14 α . Thus, the structure of metabolite **2** was defined as cholesta-5,25(27)-diene-1 β ,3 β ,7 β ,14 α ,26-pentaol-16,22-dione.

The molecular formula of metabolite **3** was established as $C_{27}H_{40}O_7$ based on ¹³C NMR and HRMS data (obsd $[M + Na]^+$, m/z 499.2656; calcd 499.2672). While the carbon resonances are similar, the ¹H NMR spectrum of **3** showed two discrepancies in comparison to **2**. When the spin-system starting from the distinguishing olefinic resonance H-6 was established, the upfield shift of the oxymethine proton H-7 was significant, suggesting a configurational difference in metabolite **3** (δ_H 4.96 in **2**, δ_H 4.32 in **3**). As a result of this configurational change, the H₂-15 resonances were also affected (δ_H 3.20 and 3.05 in **2**, δ_H 2.64 and 2.49 in **3**), signifying an intramolecular

hydrogen bond between the C-14 and C-7 hydroxy groups. The orientational difference of C-7(OH) was further supported by the ROESY spectrum. The cross-peak from β -oriented H-8 ($\delta_{\rm H}$ 1.95) to H-7 ($\delta_{\rm H}$ 4.32) and from α -oriented H-9 ($\delta_{\rm H}$ 2.77) to the exchangeable proton of C-14(OH) ($\delta_{\rm H}$ 6.48) confirmed α cofacial orientations of C-7(OH) and C-14(OH). On the basis of these findings, the structure of compound **3** was defined as cholesta-5,25(27)-diene-1 β ,3 β ,7 α ,14 α ,26-pentaol-16,22-dione.

The HRMS spectrum of metabolite 4 showed a major ion peak at m/z 497.2532 (calcd for $C_{27}H_{38}O_7Na$ 497.2512), indicating a molecular formula of $C_{27}H_{38}O_7$ with nine indices of hydrogen deficiency. The ¹H and ¹³C NMR spectra of 4 were similar to those of 2 and 3, except the signals deriving from an additional carbonyl resonance (δ_C 200.1) and those in its close proximity. Examination of the 1D and 2D NMR spectra of 4 permitted establishment of the additional carbonyl group at C-7. The correlation of H-15_{av} which was deduced by the crosspeak from H-15_{ax} to Me-18 β , and exchangeable proton of C-14(OH) (δ_H 6.39) in the ROESY spectrum revealed the orientation of the hydroxy group at C-14 to be α . Consequently, the structure of metabolite 4 was defined as cholesta-5,25(27)-diene-1 β ,3 β ,14 α ,26-tetraol-7,16,22-trione.

The molecular formula of **5** was determined as $C_{29}H_{40}O_8$ based on the major ion peak at m/z 555.2377 [M + K]⁺ (calcd

Table 3. ¹H and ¹³C NMR Data of Compounds 12–15 (500 MHz, Pyridine-d₅)

		12		13		14		15
position	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{\rm C}$	$\delta_{ m H}~(J~{ m in~Hz})$	$\delta_{\rm C}$	$\delta_{ m H}$ (J in Hz)
1	75.3	3.93 m	78.3	3.82 d (10.5)	76.2	3.84 m	77.5	4.30 m
2	41.7	2.23 m, 2.70 m	42.4	2.35 m, 2.57 m	43.2	2.25 m, 2.60 m	41.4	2.31 m, 2.58 m
3	67.4	3.88 m	67.0	3.91 m	65.9	3.90 m	65.2	4.17 m
4	43.9	2.70 m	43.6	2.70 m, 2.80 m	43.6	2.75 m	43.4	2.77 m
5	143.2		164.7		165.3		63.4	
6	128.7	6.07 d (5.5)	126.2	6.12 s	128.4	6.02 s	64.8	3.41 d (3.2)
7	64.3	4.11 bs	200.8		200.9		68.1	4.75 m
8	38.6	1.77 m	79.0		49.5	2.94 m	36.6	2.47 m
9	49.3	2.32 dd (10.0, 10.5)	47.1	3.18 d (3.5)	45.3	2.84 m	36.3	2.88 m
10	46.4		45.7		44.9		40.8	
11	66.2	4.38 d (12.5)	21.0	2.47 ddd (13.5, 13.0, 12.5) 3.25 d (13.0)	23.2	1.99 d (12.5), 2.88 m	24.5	1.76 m, 2.57 m
12	49.7	1.70 t (11.5), 2.51 m	33.7	1.58 d (12.0), 2.38 dd (12.0, 14.5)	31.9	1.44 m, 2.33 m	32.7	1.36 m, 2.33 m
13	40.1		45.6		45.4		44.7	
14	48.6	1.92 m	86.5		84.5		88.3	
15	31.8	1.57 m, 2.51 m	37.6	2.71 m, 3.42 ddd (7.0, 7.5, 7.5)	41.9	2.24 ddd (11.5, 12.0, 12.5), 3.53 ddd (7.5, 7.5, 5.5)	40.5	1.92 m, 2.45 m
16	81.6	4.58 ddd (7.5, 7.5, 7.0)	82.2	5.14 m	82.3	5.08 m	82.1	5.06 m
17	62.3	1.93 dd (7.5, 8.0)	58.8	2.82 m	58.2	2.71 m	59.1	2.69 m
18	16.8	0.95 s	21.0	1.74 s	20.3	1.13 s	20.0	1.04 s
19	12.3	1.41 s	14.3	1.90 s	12.1	1.46 s	11.1	1.55 s
20	41.7	1.97 dd (7.5, 7.5)	40.6	2.17 m	41.7	2.23 m	41.5	2.04 dd (7.0, 8.0)
21	14.7	1.01 d (6.5)	15.2	1.12 d (7.0)	15.2	1.11 d (7.0)	20.5	1.05 d (9.0)
22	109.3		109.6		109.6		109.5	
23	32.9	1.76 m	33.1	1.76 m, 1.80 m	33.1	1.79 m	33.1	1.78 m
24	28.8	2.23 m, 2.70 m	28.7	2.15 m, 2.60 m	29.5	2.21 m, 2.70 m	28.8	2.25 m, 2.70 m
25	144.7		144.3		144.5		144.3	
26	64.9	3.98 d (12.5), 4.38 d (12.5)	64.6	3.96 d (12.5), 4.40 d (12.0)	64.7	4.41 d (12.0), 3.95 d (12.0),	64.7	4.43 d (12.0), 3.99 d (12.0)
27	109.3	4.79 s, 4.77 s	108.4	4.71 s, 4.74 s	108.4	4.75 d (8.5)	108.5	4.77 d (10.0)

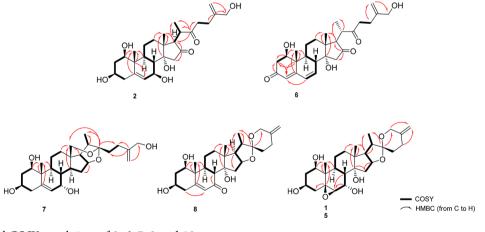


Figure 3. HMBC and COSY correlations of 2, 6, 7, 8, and 15.

555.2360 for C₂₉H₄₀O₈K) in the HRMS spectrum. In the ¹H and ¹³C NMR spectra of **5**, the resonances were almost superimposable with those of compound **4**, except additional signals originating from an acetyl group [$\delta_{\rm H}$ 2.00, s; $\delta_{\rm C}$ 20.5 (q) and 170.4 (s)]. In the ¹H NMR spectrum, the downfield shift for H₂-26 (ca. 0.30 ppm at $\delta_{\rm H}$ 4.69) and an upfield shift for H₂-27 (ca. 0.30 ppm at $\delta_{\rm H}$ 5.13) implied the acylation position as C-26(O). This assumption was substantiated by the long-range HMBC correlation from the ester carbonyl signal at $\delta_{\rm C}$ 170.4 to H₂-26a ($\delta_{\rm H}$ 4.69). Based on the NOE correlations from H-9

and H-17 ($\delta_{\rm H}$ 2.94 and 3.87, respectively) to the exchangeable proton, the orientation of C-14(OH) ($\delta_{\rm H}$ 6.41) was determined to be α . Thus, the structure of **5** was elucidated as 26-O-acetylcholesta-5,25(27)-diene-1 β ,3 β ,14 α ,26-tetraol-7,16,22-trione.

The HRESIMS spectrum of 6, exhibited a sodium adduct ion at m/z 479.2403 [M + Na]⁺ (calcd 479.2410) and led to establishment of its molecular formula as $C_{27}H_{36}O_6$ with 10 indices of hydrogen deficiency. In the ¹H and ¹³C NMR spectra of 6, the signals arising from the rings C and D and the side

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chain were similar to those of compounds 2-5, accounting for five indices of hydrogen deficiency. Inspection of the 1D and 2D NMR spectra revealed an additional carbonyl group (δ_C 197.7) together with a trisubstituted ($\delta_{\rm C}$ 124.3, 162.7, and $\delta_{\rm H}$ 5.93) and a disubstituted ($\delta_{\rm C}$ 130.1, 140.1; $\delta_{\rm H}$, 6.30, 6.25, respectively) double-bond system. The oxymethine resonance at $\delta_{\rm C}$ 72.5 was readily assigned to C-1 based on its long-range correlation with H₃-19 ($\delta_{\rm H}$ 1.40). In the COSY spectrum, the H-1 ($\delta_{\rm H}$ 4.27) resonance was correlating with a methylene group ($\delta_{\rm H}$ 2.99 and 3.05, H₂-2), which in turn had no further cross-peaks. In the HMBC spectrum, the ${}^{2}J_{C-H}$ correlation between the carbonyl signal at 197.7 ppm (C-3) and $\delta_{\rm H}$ 2.99/ 3.05 (H₂-2) protons verified an oxidation at C-3 to give a 3carbonyl group in 6. The trisubstituted olefinic system was located at C-4(5) ($\delta_{\rm C}$ 124.3 and 162.7, respectively) on the basis of the deshielding of C-5 due to the α,β -unsaturated carbonyl functionality and key HMBCs from C-2 ($\delta_{\rm C}$ 44.9) and C-10 ($\delta_{\rm C}$ 42.4) to H-4 ($\delta_{\rm H}$ 5.93) and from C-5 ($\delta_{\rm C}$ 162.8) to H_3 -19 (δ_H 1.40). Thus, the disubstituted double-bond resonances ($\delta_{\rm H}$ 6.25, dd, J = 9.5, 1.0 Hz; $\delta_{\rm H}$ 6.31, dd, J = 9.5, 2.0 Hz; $\delta_{\rm C}$ 130.1 and 140.1, respectively) were directly assigned to C(H)-6 and C(H)-7 (Figure 3). Consequently, the structure of 6 was determined as cholesta-4,6,25(27)-triene-1 β ,14 α ,26triol-3,16,22-trione.

The HRMS spectrum of metabolite 7 provided a sodiated ion peak at m/z 483.2784 $[M + Na]^+$ (calcd 483.2723 for $C_{27}H_{40}O_6Na$), indicating eight indices of hydrogen deficiency and a 32 amu increase over 1. In the ¹H NMR spectrum of 7, the resonance of one of the tertiary methyl groups was absent, and an isolated oxymethylene group was evident from the ¹H and ¹³C NMR and HSQC spectra as an AB system ($\delta_{\rm H}$ 4.07, d, J = 11.9 Hz; $\delta_{\rm H}$ 3.41, d, J = 11.8 Hz; $\delta_{\rm C}$ 65.2, t). Examination of the COSY and TOCSY spectrum starting from the characteristic resonances, namely, oxymethine proton H-1 and olefinic H-6, permitted the location of the two major spin systems in the steroid framework (Figure 3) and revealed an oxygenation at C-7 ($\delta_{\rm C}$ 64.3; $\delta_{\rm H}$ 4.07) as in metabolites 2 and 3. Furthermore, the new oxymethylene group was attributed to CH₂-18 based on the key HMBCs from C-17 and C-14 to H₂-18 (δ 4.07 and 3.41). Interestingly, in the COSY spectrum, H-17 did not couple with H-20, which in turn correlated with CH₃-21, suggesting a ca. 90° dihedral angle between H-17 and H-20. In addition, the third spin system was tracked from the characteristic H₂-27 signals ($\delta_{\rm H}$ 5.04 and 5.38) through allylic couplings to the hydroxymethyl H₂-26 ($\delta_{\rm H}$ 4.43, 2H) and H₂-24 ($\delta_{\rm H}$ 2.03 and 2.23); the latter in turn coupled with H₂-23 ($\delta_{\rm H}$ 2.47 and 2.61). The diastereotopic nature of H₂-26, resonating as a 2H broad singlet, and carbon chemical shifts of C-25 ($\delta_{\rm C}$ 151.1) and C-27 ($\delta_{\rm C}$ 107.8) demonstrated cleavage of the C-22/C-26 oxygen bridge to afford an acyclic side chain. The carbon resonance observed at $\delta_{\rm C}$ 109.9 was assigned to C-22 based on the ${}^{3}J_{C-H}$ correlations from this carbon to H-16, H-17, H₃-21, and H₂-24 in the HMBC spectrum. The correlation between H-16 and C-22 verified the oxygen bridge (Figure 3), suggesting a furostanol-type framework for metabolite 7. While all of the structural components accounted for seven out of eight indices of hydrogen deficiency, the absence of an additional multiple bond in the structure implied a recyclization. This assumption was substantiated by the longrange HMBC correlation between one of the H₂-18 protons $(\delta_{\rm H}$ 4.07) and C-22 $(\delta_{\rm C}$ 109.9), revealing epoxy bridge formation and thus the acetal character of C-22. This sixmembered ring system, which forced the H-16/C-16/C-20/H-

20 system to adopt a dihedral angle of 82° [see MM2-minimized energy conformer of 7 (Figure 4)], also explained

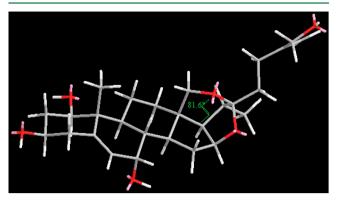


Figure 4. MM2 energy minimized conformer of 7.

the absence of coupling between H-17 and H-21. The α orientation of the hydroxy group at C-7 was deduced via the correlation of β -oriented H-8 ($\delta_{\rm H}$ 1.77) and H-7 ($\delta_{\rm H}$ 4.07) in the NOESY spectrum. Thus, the structure of metabolite 7 was established as 18(22)-epoxyfurosta-5,25(27)-diene-1 β ,3 β ,7 α ,26-tetraol.

The HRMS data of metabolite 8 exhibited a protonated molecular ion at m/z 445.2956 $[M + H]^+$ (calcd for $C_{27}H_{41}O_5$ 445.2954), revealing a 16 amu increase over NR (1). In the low-field region of the ¹H NMR spectrum, an additional low-field proton at $\delta_{\rm H}$ 4.09 was observed. Furthermore, deshielding of the characteristic olefinic resonance of H-6 ($\delta_{\rm H}$ 6.06, d, J = 5.0 Hz) inferred an adjacent oxygenation. The significant COSY correlation between the resonance at $\delta_{\rm H}$ 4.09 (H-7) and H-6 ($\delta_{\rm H}$ 6.06) implied the position of functionalization as C-7 (Figure 3). The long-range correlations of C-5 ($\delta_{\rm C}$ 143.1) and C-6 ($\delta_{\rm C}$ 128.0) with this oxymethine proton also substantiated the proposed modification. The α -orientation of C-7(OH) was established via the ROESY cross-peak between H-8 ($\delta_{\rm H}$ 1.77) and H-7. Based on these findings, the structure of 8 was determined as spirost-5,25(27)-diene-1 β ,3 β ,7 β -triol.

In the HRMS spectrum of 9, a sodium adduct ion was observed at m/z 483.2733 [M + Na]⁺ (calcd for C₂₇H₄₀O₆Na 483.2723), signifying the presence of two hydroxy groups due to the 32 amu increase compared to 1. Inspection of the ¹H NMR spectrum of 9 revealed an additional low-field signal at $\delta_{\rm H}$ 4.46, whereas a new oxymethine carbon resonance at $\delta_{\rm C}$ 66.3 was noted in the ¹³C NMR and DEPT135 spectra together with an oxygenated tertiary carbon at $\delta_{\rm C}$ 87.5. The correlation between H-6 ($\delta_{\rm H}$ 5.94, d, J = 5.0 Hz) and the new signal resonating at $\delta_{\rm H}$ 4.46 verified the first oxygenation at C-7, as in compound 8. The long-distance correlations between the tertiary carbon at $\delta_{\rm C}$ 87.5 and H₃-18 ($\delta_{\rm H}$ 1.12), H-8 ($\delta_{\rm H}$ 2.02), and H₂-15a ($\delta_{\rm H}$ 2.62) in the HMBC spectrum verified an oxygenation at C-14. In the ROESY spectrum, correlation of H-7 ($\delta_{\rm H}$ 4.46) with β -oriented H-8 ($\delta_{\rm H}$ 2.02) and a cross-peak between the exchangeable proton signal of 14-OH at $\delta_{\rm H}$ 6.18 and H-17 α ($\delta_{\rm H}$ 2.80) revealed the α -orientations of the hydroxy groups. Consequently, the structure of metabolite 9 was defined as spirosta-5,25(27)-diene- 1β , 3β , 7α , 14α -tetraol.

The molecular formula of **10** was determined to be $C_{27}H_{40}O_7$ based on the protonated molecular ion at m/z 477.2848 [M + H]⁺ (calcd for $C_{27}H_{41}O_7$ 477.2858), and the 48 amu difference in comparison to **1** suggested a trihydroxy analogue. Examination of the 1D and 2D NMR spectra revealed that the first two oxygenation positions were C-7 and C-14 as in 9. The remaining oxymethine resonance at $\delta_{\rm C}$ 72.4 displayed a strong ${}^{3}J_{\rm C-H}$ correlation with the methyl signal of H₃-18 ($\delta_{\rm H}$ 1.37), substantiating the other oxygenation position at C-12. From the ROESY spectrum, the α -oriented hydroxy group at C-7 was readily deduced via the correlation of β -oriented H-8 ($\delta_{\rm H}$ 2.07) and H-7 ($\delta_{\rm H}$ 4.52), whereas C-12(OH) was assigned a β -orientation due to the cross-peak between H-17 α ($\delta_{\rm H}$ 3.18) and H-12 ($\delta_{\rm H}$ 4.96). Furthermore, the orientation of C-14(OH) was inferred from the correlation of the exchangeable proton ($\delta_{\rm H}$ 6.16) with H-15a ($\delta_{\rm H}$ 2.66), which in turn had a correlation with H-16 α ($\delta_{\rm H}$ 5.22, ddd, J = 7.5, 7.5, 6.0 Hz). Consequently, the structure of **10** was elucidated as spirost-5,25(27)-diene-1 $\beta_3\beta_7\pi_a$,12 β_1 4 α -pentaol.

The HRMS spectrum of metabolite 11 provided a protonated molecular ion at m/z 461.2904 $[M + H]^+$ (calcd for $C_{27}H_{41}O_6$ 461.2903), indicating a 32 amu increase over NR (1). Two additional low-field oxymethine resonances at $\delta_{\rm C}$ 79.2 and 64.7 deduced from the ¹³C NMR and DEPT135 spectra suggested two sites of oxygenation. The first location was evident from the COSY correlation from the olefinic H-6 ($\delta_{\rm H}$ 6.07) to H-7 ($\delta_{\rm H}$ 4.14), whereas the key long-range correlation between the new oxymethine carbon resonance at $\delta_{\rm C}$ 79.2 and characteristic H₃-18 ($\delta_{\rm H}$ 1.24) together with the spin system tracked from H-6 to H-12 justified the position of the second hydroxy group at C-12 as in 10. The ROESY correlation of H-12 with H-17 α ($\delta_{\rm H}$ 2.25) and a cross-peak between H-7 ($\delta_{\rm H}$ 4.14) and H-8 β ($\delta_{\rm H}$ 1.77) revealed the orientations of the OH groups as β and α , respectively. Thus, the structure of metabolite 11 was elucidated as spirosta-5,25(27)diene- 1β , 3β , 7α , 12β -tetraol.

The HRMS data of metabolite 12 exhibited a protonated molecular ion at $m/z [M + H]^+$ 461.2901 (calcd for C₂₇H₄₁O₆ 461.2903), revealing a 32 amu increase over 1. Together with the characteristic signals of NR (1) an additional low-field resonance at $\delta_{\rm H}$ 4.11 was observed in the ¹H NMR spectrum, whereas ¹³C NMR and DEPT135 spectra revealed two additional oxymethine signals, suggesting a dihydroxy analogue of **1**. From the characteristic methyl doublet (H₃-21: $\delta_{\rm H}$ 1.01, d, J = 6.5 Hz) and H-6 olefinic resonance ($\delta_{\rm H}$ 6.07, d, J = 5.5) (Table 3), the spin system and its corresponding carbons constituting the B, C, D, and E rings were deduced using the COSY, TOCSY, and HSQC spectra and revealed oxygenation sites as C-7 and C-11. The key HMBCs from C-11 ($\delta_{\rm C}$ 66.2) to H-12a ($\delta_{\rm H}$ 2.51) and H-9 ($\delta_{\rm H}$ 2.32, dd, *J* = 10.0, 10.5) and from C-6 ($\delta_{\rm C}$ 128.7) and C-8 ($\delta_{\rm C}$ 38.6) to H-7 ($\delta_{\rm H}$ 4.11) also supported these findings. Observation of the cross-peaks between H-11 ($\delta_{\rm H}$ 4.38, d, J = 12.5 Hz) and H₃-18 β ($\delta_{\rm H}$ 0.95) and H_3-19 ($\delta_{\rm H}$ 1.41) and between H-7 ($\delta_{\rm H}$ 4.11) to H-8 $(\delta_{\rm H} 1.77)$ substantiated α -cofacial orientations of both HO-11 and HO-12. Based on these findings, the structure of 12 was defined as spirosta-5,25(27)-diene- 1β , 3β , 7α , 11α -tetraol.

The HRMS spectrum of metabolite **13** provided a protonated molecular ion at m/z 475.2704 [M + H]⁺, supporting the molecular formula of $C_{27}H_{39}O_7$ (calcd for $C_{27}H_{40}O_7$ 475.2696). An α,β -unsaturated carbonyl functionality (δ_C 200.8, 164.7, and 126.2) similar to those of **4** and **5** was evident from the ¹³C NMR spectrum of **13**. This system was attributed to C-5=C-6-C-7(=O) on the basis of the long-range HMBC correlations from C-5 to H₃-19 (δ_H 1.90) and from C-4/C-10 to H-6. Other distinguishing features observed for metabolite **13** were the presence of two oxygenated tertiary carbon resonances (δ_C 79.0, 86.5) in the ¹³C NMR spectrum.

When the COSY and TOCSY spectra were inspected beginning from the H₂-12 resonances ($\delta_{\rm H}$ 1.58 and 2.38), the spin system H₂-12 \rightarrow H₂-11 ($\delta_{\rm H}$ 2.47 and 3.25) \rightarrow H-9 ($\delta_{\rm H}$ 3.18) was established, suggesting the first oxygenation at C-8 due to the absence of correlation between H-9 and H-8. Moreover, the long-range correlation from the carbon at $\delta_{\rm C}$ 79.0 to H-6 ($\delta_{\rm H}$ 6.12) confirmed oxygenation at C-8. The second hydroxy group was located at C-14 as in 9 and 10 because of the significant HMBCs from the carbon resonating at $\delta_{\rm C}$ 86.5 signal to H₃-18 ($\delta_{\rm H}$ 1.74) and H₂-15 ($\delta_{\rm H}$ 2.71 and 3.42). The exchangeable proton of C-8(OH) ($\delta_{\rm H}$ 7.41) showed ROESY cross-peaks with H₃-18 and H₃-19, revealing its β orientation, whereas the C-14(OH) proton ($\delta_{\rm H}$ 4.92) had correlations with H-9 ($\delta_{\rm H}$ 3.18) and H-17 ($\delta_{\rm H}$ 2.82), substantiating its α -orientation as in 9 and 10. Consequently, the structure of metabolite 13 was defined as spirosta-5,25(27)diene- 1β , 3β , 8β , 14α -tetraol-7-one.

In the HRMS spectrum of 14, a sodium adduct ion was observed at m/z 481.2575 [M + Na]⁺ (calcd 481.2566 for $C_{27}H_{38}O_6Na$), indicating a 30 amu difference in comparison to 1. Inspection of the ¹H and ¹³C NMR spectra of 14 suggested an α,β -unsaturated carbonyl moiety as in 13 (δ_C 165.3, 128.4, and 200.9; δ_{H-6} 6.02), which was also corroborated via key HMBC correlations. Examination of the ¹³C and DEPT135 spectra indicated an additional carbon signal at δ_C 84.5. A long-range correlation between this carbon and H₃-18 (δ_H 1.13) suggested oxygenation at C-14 as in 9, 10, and 13. Based on the ROESY correlations between C-14(OH) and H-16/H-9, the hydroxy group was α -oriented. Thus, the structure of 14 was elucidated as spirosta-5,25(27)-diene-1 $\beta,3\beta,14\alpha$ -triol-7-one.

The molecular formula of 15 was found to be $C_{27}H_{40}O_6$ based on the sodium adduct ion at m/z 483.2728 [M + Na] (calcd for C₂₇H₄₀O₆Na 483.2723), indicating eight indices of hydrogen deficiency. In the low-field region of the ¹H and ¹³C NMR spectra, the absence of characteristic resonances (C-5, C-6 and H-6) derived from the olefinic system was noted. Based on the absence of unsaturation in 15, an additional ring system was proposed based on the identical hydrogen deficiency index compared to NR (1). The major spin systems were established by COSY and TOCSY spectra, which suggested modifications in rings B and C. The oxygenation at C-14 was readily inferred based on the ${}^{3}J_{C-H}$ HMBC correlation between C-14 and H₃-18 ($\delta_{\rm H}$ 1.04). In the COSY spectrum, the last spin system was tracked from the characteristic H₂-12 ($\delta_{\rm H}$ 1.36 and 2.33) to the oxymethine H-7 resonance ($\delta_{\rm H}$ 4.75). The C-7 hydrogen also coupled with another low-field signal resonating at $\delta_{\rm H}$ 3.41, readily assigned to H-6, hence establishing the spin system from H₂-12 to H-6. In the COSY spectrum, H-7 showed a correlation with an exchangeable proton at $\delta_{\rm H}$ 7.92, confirming the second oxygenation position in the structure (Figure 3). The carbon observed at $\delta_{\rm C}$ 63.4 was ascribed to C-5 on the basis of its key HMBCs to H₃-19 ($\delta_{\rm H}$ 1.55) and H₂-4 ($\delta_{\rm H}$ 2.77, 2H, m). When the high-field shift of C-5 was evaluated together with the remaining index of hydrogen deficiency, the presence of an epoxide moiety was evident between C-5 and C-6.^{31,32} Based on the NOESY cross-peaks between H-7 and H-8 β /H-15 β , C-14(OH) ($\delta_{\rm H}$ 5.61), and H-9 α ($\delta_{\rm H}$ 2.88) and the absence of correlation between HO-7 and HO-6, C-7 and C-14 (OH) were α -cofacially oriented, whereas the epoxide ring had β orientation. Thus, the structure of metabolite 15 was elucidated as $5\beta(6)$ -epoxyspirost-25(27)-ene- 1β , 3β , 7α , 14α -tetraol.

The major steroidal sapogenins of Ruscus sp., neoruscogenin (1) and ruscogenin, are responsible for vasoconstrictive,

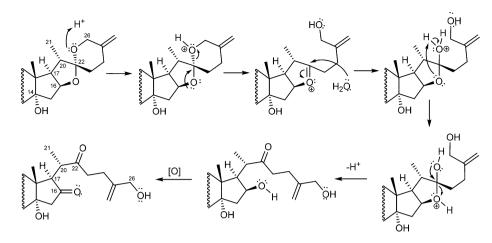


Figure 5. Putative mechanism of spiroketal ring cleavage to yield the cholestane-type framework.

venotonic, antithrombotic, and anti-inflammatory activity.^{18,19,23,33,34} The extracts of the underground part of *Ruscus* have been standardized to contain these spirostanol steroids as a mixture, in which neoruscogenin (1) is the major constituent.

Microbial transformation has been widely used to obtain pharmacologically active steroids. It is advantageous in comparison to chemical synthesis due to its capability to perform stereo- and regioselective reactions at active and inactive sites of the substrate molecules.^{4,35–37} The recent studies have shown that endophytes can be used as biocatalysts due to their ability to catalyze diverse and unique reactions.^{11–13,38,39}

Our previous studies revealed the biotransformation capacity of the filamentous fungi, viz. Cunninghamella blakesleeana and Glomerella fusarioides, toward triterpenoids.⁴⁰⁻⁴² In this study, we demonstrated the potential of an endophytic fungus (A. eureka) as biocatalyst to transform the steroidal sapogenin neoruscogenin. As a result, 14 new metabolites were obtained, and their structures were elucidated. Metabolites 8-15 were mainly transformed via actions of CYP450 monooxygenase enzymes, resulting in oxygenation reactions (8-12 and 15), whereas subsequent dehydrogenation reactions provided carbonyl derivatives (13 and 14). As it was expected from the literature, the main oxygenation site was C-7 due to its allylic nature in the B ring.^{4,43,44} The other oxygenation location was C-14, which was shown to be subject to oxygenation as in previous studies with other fungi.45 Oxygenations at C-8, C-11, and C-12 were also observed. C-11 and C-12 oxygenations are more common in steroid chemistry; however, C-8 is at an encumbered position on the steroid skeleton, and modification of this stereocenter is rare due to steric hindrance. Based on the literature, only Alternaria alternata and Corynespora cassiicada CYP450s are capable of performing this unique oxygenation.48,49

The other striking reaction occurring in compound 1 was epoxidation at the $\Delta^{5(6)}$ olefinic bond to afford 15. Even though there are several reports of microbial^{50–53} or semisynthetic^{54,55} epoxidation products of steroids, this is the first study demonstrating a $\Delta^{5(6)}$ epoxidation on a steroidal skeleton by a microbial enzyme system.

A spiroketal hydrolysis reaction of 1 afforded the acyclic side chain of metabolites 2-6 (Figure 5). Subsequently 2-6 are susceptible to further transformations on the acyclic/cholestane-type steroidal framework to give oxygenation, oxidation, acylation, migration, and dehydration products. Cleavage of the C-22(O)C-26 oxygen bridge afforded a furostanol-type skeleton, and its modification resulted in the formation of a C-18(O)C-22 epoxide ring system (7). These cleavages and further epoxide ring formation are reported for the first time in a microbial transformation study. In addition, a spiroketal hydrolysis reaction is important for the synthesis of potent cytotoxic metabolites under mild conditions, viz., the cholestane-type steroid OSW-1 (3β ,16 β ,17 α -trihydroxycholest-5-en-22-one 16-O-(2-O-4-methoxybenzoyl- β -D-xylopyrano-syl)-(1 \rightarrow 3)-(2-O-acetyl- α -L-arabinopyranoside)).^{56–59}

In summary, the catalytic diversity of the endophytic fungus *A. eureka* is remarkable. These results prove the capability of endophytes for catalyzing novel reactions and warrant further studies to exploit their use in the field of natural product and bioorganic chemistry. Finally, as can be seen from the diversity of the metabolites, neoruscogenin (1), the major bioactive component of the *Ruscus* preparations, is subject to many biotransformation reactions, which might be taken into consideration for its conceivable mammalian metabolites, as well as its current or potential therapeutic use.

EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were measured using a PerkinElmer 341 polarimeter. MS data were acquired on an Agilent 1200/6530 Instrument–HRTOFMS. 1D and 2D NMR spectra were obtained at 500 MHz for ¹H and 125 MHz for ¹³C on a Bruker DRX 500 spectrometer, in pyridine- d_5 , with solvent peaks used as references. Column chromatography experiments were carried out on silica gel 60 (40–63 μ m-Merck), Sephadex LH-20 (GE Healthcare), and RP (C₁₈, 25–40 μ m) (Merck). TLC analyses were carried out on silica gel 60 F254 (Merck) and RP-18 F254s (Merck) plates. HPTLC experiments were carried out on HPTLC silica gel 60 F254 (Merck). Compounds were detected by UV and 20% aqueous H₂SO₄ spraying reagent followed by heating at 105 °C for 1–2 min.

Substrate. The substrate neoruscogenin was isolated and purified from *Ruscus aculeatus* L., donated by Bionorm Natural Products Production & Marketing Corp., Menderes-Izmir, Turkey. Its purity was determined to be 95% by HPLC analysis.

Fungal Strain and Culture Media. The endophytic fungus *Alternaria eureka* was isolated from *Astragalus angustifolius*, collected in Manisa-Spil Mountain, Izmir, Turkey, in June 2013. It was identified based on DNA sequence analysis conducted by DSMZ (Leibniz Institute DSMZ-German Collection of Microorganisms and Cell Cultures). The original culture was deposited at the Erdal Bedir Laboratory with the deposit number 20131E1BL1.⁴¹ All cultures were maintained on potato dextrose agar (PDA) slants and stored at 4 °C

until use. Prior to biotransformation, the fungus was precultivated on PDA in Petri dishes for 5-7 days at 25 °C.

Microbial Transformation Procedures. The biotransformation process was conducted at two scales, analytical and preparative. Inoculum of 2% derived from the suspension of 5-7-day-old cultures with Tween 80 (0.1%) was used in the biotransformation process. A one-stage fermentation protocol was followed, where 1 was fed to the media 72 h after the inoculation. The biotransformation media contained 2% glucose, 0.5% yeast extract, 0.5% NaCl, 0.5% K₂HPO₄, and 0.5% (w/v) peptone (pH 6.0). Analytical scale was conducted using 250 mL flasks containing 50 mL of media and 1 (10 mg). Every 2 days, samples with a volume of 0.5 mL were taken and centrifuged. The supernatants were extracted with EtOAc. In order to differentiate the transformed products from the metabolites of the microorganisms, one flask was kept under the same conditions but without lead compound. In preparative scale, 34 Erlenmeyer flasks each containing 200 mL of biotransformation media and 40 mg of 1 were used at the same conditions as analytical scale (25 °C and 180 rpm) for 10 days.

Extraction and Isolation. After the incubation period, prior to the extraction with EtOAc (×3) and n-BuOH, microorganisms were separated from the media by using a Buchner funnel. After drying over anhydrous Na2SO4, the organic layers were evaporated under reduced pressure. The EtOAc (1.373 g) and n-BuOH (1.253 g) extracts were applied to Sephadex LH-20 column chromatography (CC) (100 g, 3.5 \times 43 cm) separately and eluted with MeOH. The fractions from each column with NR metabolites, viz. 19-28 (1.157 g) and 19 and 20 (284.9 mg), were pooled and subjected to reversed-phase silica gel $(C_{18}, 70 \text{ g}, 3.5 \times 26 \text{ cm})$ to yield 39 main fractions after elution with a MeOH/H2O gradient (30:70 100 mL, 35:65 100 mL, 40:60 100 mL, 45:55 100 mL, 50:50 300 mL, 55:45 200 mL, 60:40 200 mL, 70:30 300 mL, 80:20 300 mL, 90:10 250 mL, 100:0 250 mL). Fraction A (33.7 mg) was purified on silica gel open-column chromatography (8 g; 2 \times 9.5 cm), eluting with CHCl₃/MeOH/H₂O (90:10:0.5), to afford 3.5 mg of 2. In order to isolate metabolites 4 (5.1 mg) and 7 (4.6 mg), fraction B (42.3 mg) was subjected to silica gel open-column chromatography (12 g; 2 \times 13 cm) and eluted with $\tilde{C}H\tilde{C}l_3/MeOH/$ H₂O (95:5:0.5). Fraction C (63.0 mg) was first purified on silica gel CC (15 g, 2×14 cm) with CHCl₃/MeOH/H₂O (95:5:0.5) to give 6 (1.5 mg). To isolate metabolite 3 (0.5 mg), subfraction C, C1 (5.5 mg), was subjected to preparative TLC (HPTLC) eluting with nhexane/EtOAc/MeOH (10:10:4). To purify metabolite 5, fraction D (38 mg) was subjected to silica gel CC (12 g, 1.5×18 cm) and eluted with CHCl₃/MeOH (95:5). Subfraction D1 (5.5 mg) was subjected to HPTLC using CHCl₃/MeOH/H₂O (85:15:0.5). Fractions E and F1 containing the metabolite 10 (9.4 mg) were separately subjected to silica gel CC. Fraction E (30 mg) was subjected to silica gel CC (10 g, 1.5×15 cm) and eluted with CHCl₃/MeOH/H₂O (90:10:0.5). Fraction F1 (36.3 mg), the supernatant of fraction F (48.2 mg), was also subjected to silica gel CC (15 g, 1.5×20 cm) eluted with CHCl₃/ MeOH (90:10) to which 0.2 µL of HOAc was added for each 100 mL. Eluting fraction G with cyclohexane/EtOAc/MeOH (10:10:1) on silica gel CC (8 g, 1.5×12 cm) afforded 8.0 mg of metabolite 11. With further purification of subfraction G (fraction G1) (7.5 mg) on HPTLC with CHCl₃/MeOH/H₂O (90:10:0.5), 3.0 mg of metabolite 13 was obtained. Fraction H (30.5 mg) was subjected to normal-phase silica gel CC (10 g, 1.5×15) eluted with CHCl₃/MeOH (97:3) to give metabolites 12 (0.5 mg) and 15 (2.0 mg). Fractions I (55.0 mg) and J (32.0 mg) on precipitation with MeOH gave metabolites 14 (7.0 mg) and 8 (7.3 mg), respectively.

Cholesta-5,25(27)-diene- 1β , 3β , 7β , 14α ,26-pentaol-16,22-dione (2): white, amorphous solid; $[\alpha]^{25}_{\rm D}$ -48 (c 0.1, MeOH); ¹H NMR data (pyridine- d_s , 500 MHz) see Table 1; ¹³C NMR data (pyridine- d_s , 125 MHz) see Table 1; HRESIMS (positive ion mode) m/z 499.2653 $[M + Na]^+$ (calcd 499.2671 for C₂₇H₄₀O₇Na).

Cholesta-5,25(27)-diene-1 $\beta_3\beta_7/\alpha_114\alpha_26$ -pentaol-16,22-dione (3): white, amorphous solid; $[\alpha]^{25}_D$ –192 (*c* 0.02, MeOH); ¹H NMR data (pyridine-*d*₅, 500 MHz) see Table 1; ¹³C NMR data (pyridine-*d*₅, 125 MHz) see Table 1; HRESIMS (positive ion mode) *m*/*z* 499.2656 [M + Na]⁺ (calcd 499.2672 for C₂₇H₄₀O₇Na). Cholesta-5,25(27)-diene-1 β ,3 β ,14 α ,26-tetraol-7,16,22-trione (4): white, amorphous solid; $[\alpha]^{25}_{D}$ -21 (*c* 1, MeOH); ¹H NMR data (pyridine-*d*₅, 500 MHz) see Table 1; ¹³C NMR data (pyridine-*d*₅, 125 MHz) see Table 1; HRESIMS (positive ion mode) *m*/*z* 497.2532 (calcd for C₂₇H₃₈O₇Na 497.2512).

26-O-Acetylcholesta-5,25(27)-diene-1 β ,3 β ,14 α ,26-tetraol-7,16,22-trione (5): white, amorphous solid; $[\alpha]^{25}_{D}$ –191 (c 0.03, MeOH); ¹H NMR data (pyridine- d_s , 500 MHz) see Table 1; ¹³C NMR data (pyridine- d_s , 125 MHz) see Table 1; HRESIMS (positive ion mode) m/z 555.2377 [M + K]⁺ (calcd for 555.2360 C₂₉H₄₀0₈K).

Cholesta-4,6,25(27)-triene-1 β ,14 α ,26-triol-3,16,22-trione (6): white, amorphous solid; $[\alpha]^{25}_{D}$ –121 (*c* 0.1 MeOH); ¹H NMR data (pyridine- d_{5} , 500 MHz) see Table 1; ¹³C NMR data (pyridine- d_{5} , 125 MHz) see Table 1; HRESIMS (positive ion mode) *m*/*z* 479.2403 [M + Na]⁺ (calcd 479.2410 for C₂₇H₃₆O₆Na).

18(22)-Epoxyfurosta-5,25(27)-diene-1β,3β,7α,26-tetraol (7): white, amorphous solid; $[\alpha]^{25}_{D}$ –46 (*c* 0.1, MeOH); ¹H NMR data (pyridine-*d*₅, 500 MHz) see Table 2; ¹³C NMR data (pyridine-*d*₅, 125 MHz) see Table 2; HRESIMS (positive ion mode) *m*/*z* 483.2784 [M + Na]⁺ (calcd for C₂₇H₄₀O₆Na 483.2723).

Spirosta-5,25(27)-diene-1β,3β,7β-triol (8): white, amorphous solid; $[\alpha]^{25}_{D}$ –126 (*c* 0.1, MeOH); ¹H NMR data (pyridine-*d*₅, 500 MHz) see Table 2; ¹³C NMR data (pyridine-*d*₅, 125 MHz) see Table 2; HRESIMS (positive ion mode) *m*/*z* 445.2956 [M + H]⁺ (calcd for C₂₇H₄₁O₅ 445.2954).

Spirosta-5,25(27)-diene-1 β ,3 β ,7 α ,14 α -tetraol (9): white, amorphous solid; $[\alpha]^{25}_{D}$ –113 (c 0.1, MeOH); ¹H NMR data (pyridine- d_{5} , 500 MHz) see Table 2; ¹³C NMR data (pyridine- d_{5} , 125 MHz) see Table 2; HRESIMS (positive ion mode) m/z 483.2733 [M + Na]⁺ (calcd for C₂₇H₄₀O₆Na 483.2723).

Spirosta-5,25(27)-diene-1\beta,3\beta,7\alpha,12\beta,14\alpha-pentaol (10): white, amorphous solid; [\alpha]^{25}_{D} –68 (c 0.1, MeOH); ¹H NMR data (pyridine-d_{5}, 500 MHz) see Table 2; ¹³C NMR data (pyridine-d_{5}, 125 MHz) see Table 2; HRESIMS (positive ion mode) m/z 477.2848 [M + H]⁺ (calcd for C₂₇H₄₁O₇ 477.2852).

Spirosta-5,25(27)diene-1\beta,3\beta,7\alpha,12\beta-tetraol (11): white, amorphous solid; [\alpha]^{25}_{D} –161 (<i>c 0.03, MeOH); ¹H NMR data (pyridine- d_5 , 500 MHz) see Table 2; ¹³C NMR data (pyridine- d_5 , 125 MHz) see Table 2; HRESIMS (positive ion mode) *m*/*z* 461.2904 [M + H]⁺ (calcd for C₂₇H₄₁O₆ 461. 2903).

Spirosta-5,25(27)-diene-1 β ,3 β ,7 α ,11 α -tetraol (12): white, amorphous solid; $[\alpha]^{25}_{D}$ –177 (*c* 0.03, MeOH); ¹H NMR data (pyridine- d_5 , 500 MHz) see Table 3; ¹³C NMR data (pyridine- d_5 , 125 MHz) see Table 3; HRESIMS (positive ion mode) *m*/*z* 461.2901 [M + H]⁺ (calcd for C₂₇H₄₁O₆ 461. 2903).

Spirosta-5,25(27)-diene-1 β ,3 β ,8 β ,14 α -tetraol-7-one (13): white, amorphous solid; $[\alpha]^{25}_{D}$ -75 (c 0.04, MeOH); ¹H NMR data (pyridine- d_{5} , 500 MHz) see Table 3; ¹³C NMR data (pyridine- d_{5} , 125 MHz) see Table 3; HRESIMS (positive ion mode) m/z 475.2704 [M + H]⁺ (calcd for C₂₇H₃₉O₇ 475.2696).

Spirosta-5,25(27)-diene-1 β ,3 β ,14 α -triol-7-one (14): white, amorphous solid; $[\alpha]_{25}^{25}$ –212 (c 0.03, MeOH); ¹H NMR data (pyridine- d_5 , 500 MHz) see Table 3; ¹³C NMR data (pyridine- d_5 , 125 MHz) see Table 3; HRESIMS (positive ion mode) m/z 481.2575 [M + Na]⁺ (calcd for C₂₇H₃₈O₆Na 481.2566).

5β(6)-Epoxyspirosta-5,25(27)diene-1β,3β,7α,14α-tetraol (15): white, amorphous solid; $[\alpha]^{25}_{D}$ –23 (*c* 0.1, MeOH); ¹H NMR data (pyridine-*d*₅, 500 MHz) see Table 3; ¹³C NMR data (pyridine-*d*₅, 125 MHz) see Table 3; HRESIMS (positive ion mode) *m*/*z* 483.2728 [M + Na]⁺ (calcd for C₂₇H₄₀O₆Na 483.272).

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jnat-prod.7b00898.

HRESIMS and ¹H, ¹³C, and 2D NMR spectra of compounds 2–15 (PDF)

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Notes

The authors declare no competing financial interest.

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