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Characterization investigations during mechanical alloying and sintering of W–20 vol% SiC composites

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ABSTRACT

The effect of mechanical alloying and the sintering regime on the microstructural and the physical properties of W–SiC composites were investigated. Powder mixtures of W–20 vol.% SiC were mechanically alloyed (MA'd) using a Spex mill for 3 h, 6 h and 24 h. MA'd powders were characterized by Laser Diffraction Particle Size Analyzer, SEM and XRD investigations. MA'd W–20 vol.% SiC powder composites were sintered under inert Ar and reducing H_2 gas conditions at $1680\,^{\circ}$ C and $1770\,^{\circ}$ C for 1 h. The microstructural and mechanical characterizations of the sintered samples were carried out by scanning electron microscope (SEM) and X-ray diffraction (XRD) and Vickers Hardness analyses. The addition of SiC remarkably increases the hardness of the composites. Hardness is also increased with decreasing grain size and increasing amount of MA.

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1. Introduction

Many of the modern technologies require materials with unusual combinations of properties which cannot be met by conventional metal alloys, ceramics, and polymeric materials. Since no single material could fulfill these rising requirements in the last 20 years, an increasing interest in developing new materials has been aroused to serve the need for new materials having unique properties for special applications [1,2]. Therefore, in parallel with the development of composite materials, such material combinations which fulfill the requirements of the modern technology have been, and are yet being, extended. Composite materials are especially attractive for applications in marine, automotive, aviation and aerospace industries because of their exceptional strength and stiffness-to-density ratios and superior physical properties. As a result, using composite materials in these industries has led to improvements in the technology [3].

Among different types of composites, particle-reinforced composites provide cost-effectiveness and especially isotropic properties to the composite and can be processed using similar technology used for monolithic materials [4]. Furthermore, MMC'S with discontinuous reinforcements are also attractive for their processability into various shapes [5].

As interesting and appropriate matrix materials with their high melting point, high modulus, high resistance of thermal shock, low CTE and good high temperature strength and stiffness, in this last decade, tungsten and its alloys have received attention with a view of improving the high temperature mechanical properties [6]. As dispersion strengtheners, refractory carbide, nitride and oxide phases, such as TiC, ZrC, HfC, TiN, Y₂O₃, La₂O₃, Sm₂O₃, ThO₂, ZrO₂, etc. have been mainly used to improve the mechanical properties of tungsten and its alloys [6,1,7]. However, their content in tungsten is usually less than 2 vol.% and to date [7], unfortunately, research on tungsten matrix composites reinforced with a higher content of ceramic phase has not been reported. Furthermore, there is no literature which reports on tungsten matrix composites reinforced with SiC which is a hard and strong material with low density [8].

Among the various non-oxide ceramics that have found commercial applications, SiC is the best one. Attractive properties, such as good specific strength and Young modulus as a function of temperature, the specific stiffness, corrosion and erosion resistance have made SiC an attractive alternative to the hard metal compositions [8–10]. Furthermore, it is an important ceramic used in structural applications, such as automotive engines, cutting tools, heat exchange and mechanical seals [8,10,11].

Mechanical alloying (MA) is an advanced fabrication process by means of which ultra-fine and homogenous composite powders can be produced [12]. Even, nanocrystalline materials (with a grain size of few nanometers, usually <100 nm) are also produced by MA of powder mixtures [13]. Additionally, it has been recognized that

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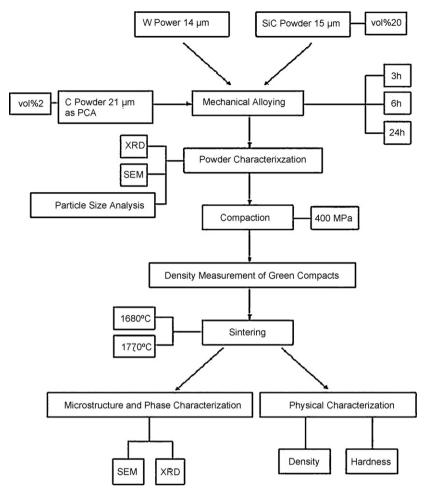


Fig. 1. The flowchart of the experimental procedure.

by using this technique chemical (displacement) reactions in powder mixtures can be induced at room temperature or at much lower temperatures than normally required to synthesize pure metals [13,14]. The structure and constitution of advanced materials can be better controlled by processing them under non-equilibrium (or far-from-equilibrium) conditions and MA is such a processing method that materials can be produced under non-equilibrium conditions [15]. MA can be defined as a powder metallurgy process for producing composite metal powders with a controlled fine microstructure by repeated cold welding, fracturing and rewelding of powder particles in a high-energy ball mill [13,16,17].

Keeping the above concepts in mind, the aim of this study has been to develop and characterize nanostructured SiC particle-reinforced tungsten composite powders and their consolidated and sintered counterparts. Composite powders with W–20 vol.% SiC composition were first investigated using SEM and XRD techniques and then mechanically alloyed in different time intervals. As-MA'd powders were consolidated using a uniaxial press and sintered under H₂ atmosphere. Bulk density and microhardness tests were carried out on sintered W–20 vol.% SiC samples.

2. Experimental procedure

Elemental tungsten (W) powders (99.9% purity, 14 μ m average particle size) as the matrix of the composite and silicon carbide (SiC) powders (99.9% purity, 15 μ m average particle size) as reinforcement were mixed to form powder batches to constitute the composition W–20 vol.% SiC (hereafter referred to as W20SiC). Furthermore, graphite powders (99.9% purity, 21 μ m average particle size) was used as a process control agent (PCA) to minimize cold welding between powder particles and thereby to inhibit agglomeration.

Mechanical alloying experiments were carried out for 3 h, 6 h and 24 h in a tungsten carbide (WC) vial using a $8000D^{TM}$ Spex mill with a speed of 1200 rpm. WC balls with a diameter of 6.35 mm (1/4 in.) were used as milling media. The ball-to-powder weight ratio (BPR) was 10:1.

Powder particle size measurements were carried out in a MalvernTM Master-sizer Laser particle size analyzer. Microstructural characterization investigations of as-blended and MA'd powders were conducted using a JeolTM-JSM-T330 scanning electron microscope (SEM), a JeolTM-JSM-7000F field emission scanning electron microscope (SEM-FEG) and a BrukerTM X-Ray Diffractometer (XRD) (Cu K_{\text{\text{\$\alpha\$}}\ radiation). Powder consolidation was carried out in a 10 ton capacity "APEXTM 3010/4" one-action hydraulic press to obtain cylinder type compacts of MA'd W2OSiC composites with a diameter of 12 mm under a uniaxial pressure of 400 MPa. As-blended}

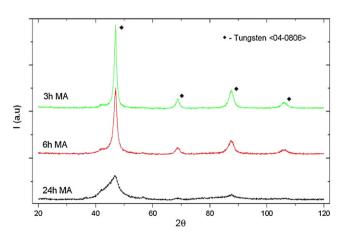


Fig. 2. XRD patterns of MA'd W20SiC samples.

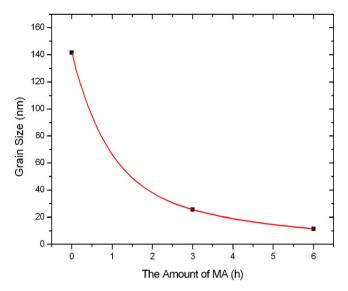


Fig. 3. Evolution of the grain size as a function of the milling time.

powders could not be compacted using these conditions and for this reason there are no pressed and sintered compacts prepared from as-blended powders used in this study. Green densities of the samples were measured dimensionally. The compacts were sintered in an Anter^IM Dilatometer at 1680 °C (1953 K) and/or in a 1800 M Vac Graphite Linn^IM gas sintering furnace at 1770 °C (2043 K) under both inert Ar and reducing $\rm H_2$ gas flowing conditions for 1 h. Microstructural characterizations of sintered W205iC samples were carried out using the same XRD and SEM equipment as those used for MA'd W20SiC powders. Because this study generated very small samples, mechanical behavior was only probed by Vickers microhardness measurements which were conducted on the sintered samples using a Shimadzu $^{\rm IM}$ microhardness tester under a load of 100 g for 15 s. Results of hardness tests were averaged out of 25 successive indentations. Sintered densities were measured by using the Archimedes method.

A Bruker™ D8 X-Ray Diffractometer was used to monitor the changes in the ball-milled powder. The crystallite size was estimated using the modified Scherrer's [18]

Table 1Weight and atomic percentages of the elements present in (a) equiaxed particles, (b) sharp-edged particles and (c) flat-like particles.

Element	Weight%	Atomic%	
(a)			
C	0.25	3.68	
Si	0.00	0.00	
W	99.75	96.32	
(b)			
C	20.32	49.12	
Si	43.72	45.20	
W	35.96	5.68	
(c)			
C	70.75	97.37	
Si	0.00	0.00	
W	29.25	2.63	

formula based on the broadening of XRD diffraction peaks:

$$B_r = \frac{k\lambda}{L\cos\theta} + \eta\tan\theta \tag{1}$$

where B_r is the total broadening due to reduction in crystallite size and lattice strains, λ the X-ray wavelength, L the crystallite size, η the strain in the material, k a constant and θ is the diffraction angle.

The experimental procedure for the present investigation is given in the flow chart in Fig. 1.

3. Results and discussion

3.1. Characterization of powders

3.1.1. Microstructural characterization

XRD diffraction patterns of mechanically alloyed W20SiC powders are shown in Fig. 2. As seen in Fig. 2, only the peaks of W which has a BCC Bravais lattice and Im3m space group with the lattice parameter of a = 0.316 nm [19] can be identified; peaks for SiC are not seen, probably due to the small amount (1.5 wt%) in

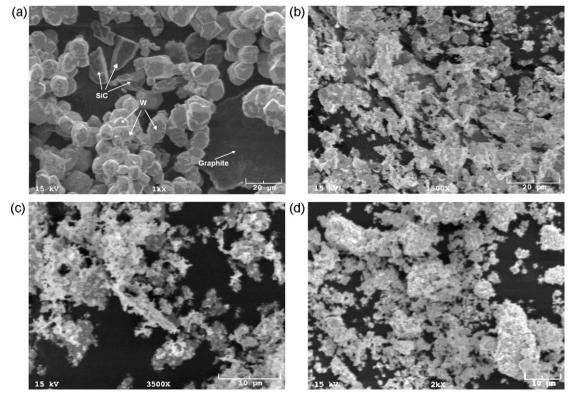


Fig. 4. SEM images of W20SiC powders (a) in the as-blended condition and after mechanical alloying for (b) 3 h, (c) 6 h, (d) 24 h.

the powder blend. Further, it is also clear that with increasing MA, peaks are broadened and peak heights are decreased, as a result of grain refinement and an increase in internal strain due to mechanical alloying. The dependence of grain size on mechanical alloying duration is shown in Fig. 3. As seen in Fig. 3, the average grain shows a rapid decrease to about 26 nm after MA'ing for 3 h and it is

about 12 nm after MA for 6 h. The grain size of the W20SiC powder MA'd for 24 h could not be estimated from the modified Scherrer's [18] equation since all the peaks except the one belonging to (110) plane disappeared due to broadening.

Fig. 4a-d shows a series of SEM micrographs taken from as-blended W20SiC powders and from those MA'd for 3 h, 6 h and

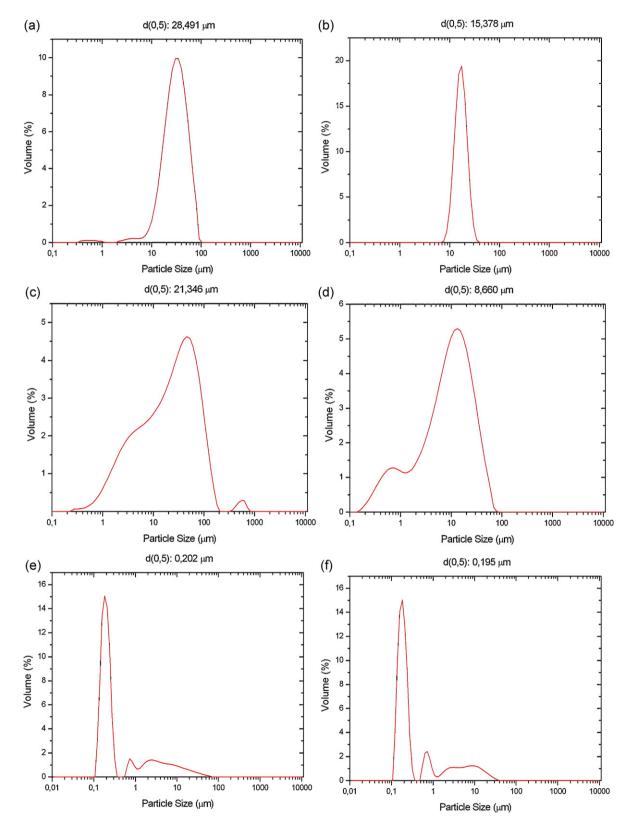


Fig. 5. Particle size distributions of (a) as-received W, (b) as-received SiC and (c) as-received graphite powders and those of W20SiC powder MA'd for: (d) 3 h, (e) 6 h, (f) 24 h.

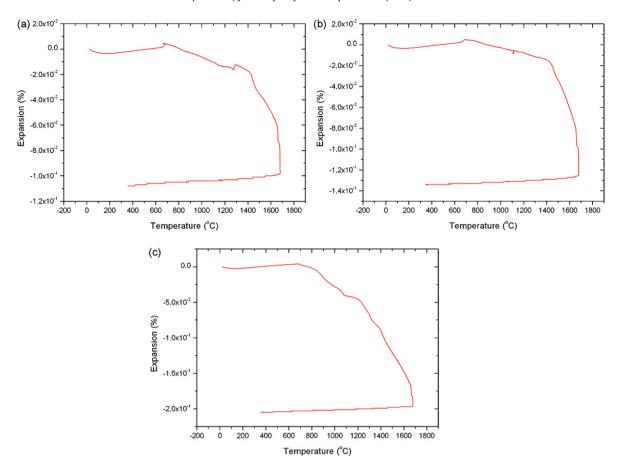


Fig. 6. Sintering curves of the W20SiC powders (a) MA'd for 3 h and sintered at 1680 °C, (b) MA'd for 6 h and sintered at 1680 °C, (c) MA'd for 24 h and sintered at 1680 °C.

24 h, respectively. It is clear from Fig. 4 that powder particle sizes decrease with increasing milling time. EDS measurements (Table 1) taken from different shaped particles seen in Fig. 4a revealed that round and nearly equiaxed particles are W particles, sharp-edged particles at the left-top of the image are SiC particles and the flake-like particle at the right-bottom of the image is graphite.

Similar to the W–SiC system of the present study, alloying has been reported to occur in brittle–brittle component systems such as Si–Ge and Mn–Bi or Fe₂O₃–Cr₂O₃ and ZrO₂–Y₂O₃ [14,15]. During MA of brittle–brittle component systems, it has been observed that the harder (more brittle) component gets fragmented and gets embedded in the softer (less brittle) component [14]. With continued milling, the more brittle particles get uniformly dispersed, if they are insoluble, in the less brittle matrix [14,16]. It can be clearly seen that in the initial stages of milling the less brittle metal (here W) powder particles get flattened by the ball–powder–ball collisions, while the brittle SiC particles get fragmented as expected [14] (Fig. 4b). With continuing milling, these flattened particles get fragmented [14] (Fig. 4c) and finally composite powder particles are formed with compositions similar to starting powder batch composition [14] (Fig. 4d).

Fig. 5a–f shows the particle size distributions of as-received W, SiC, graphite powder and those MA'd for 3 h, 6 h and 24 h, respectively. According to these analyses, the median particle size of as-received W powders is less than 28 μm (Fig. 5a), that of SiC powders is less than 14 μm (Fig. 5b) and that of graphite powders is less than 21 μm (Fig. 5c). Median particle sizes are about 8.66 μm , 0.202 μm and 0.195 μm for the W20SiC powders MA'd for 3 h, 6 h and 24 h, respectively. With increasing MA, specific surface area of the powders drastically increases, which are less than 1 m^2/g for as-blended W20SiC powders, 2.45 m^2/g for those MA'd for 3 h,

 $25.4\,\mathrm{m}^2/\mathrm{g}$ MA'd for 6 h and $26.2\,\mathrm{m}^2/\mathrm{g}$ MA'd for 24 h. It is clear from the results that the average particle sizes ($\sim\!200\,\mathrm{nm}$) and specific areas barely change between 6 h and 24 h MA times.

3.2. Characterization of as-consolidated and sintered samples

3.2.1. Determination of the sintering regime

Several factors influence the sintering regimes, including time, temperature, particle size, atmosphere, compacted density, particle size distribution, etc. [20]. For tungsten powders with an average size of 4 μm , typical sintering temperatures up to 3000 °C and times between 15 min and 30 min required to achieve high densities [21]. For the same powder, sintering times vary between 1 h and 2 h at 2400 °C and longer times (40–110 h) are required at lower temperature range between 1775 °C and 1825 °C [20,21]. However, both sintering temperatures and times can be decreased by decreasing the particle size of powders [20].

Taking these facts into account, the sintering regime is mainly determined by several additional experiments using a dilatometer, in which different flow rates and flow durations were applied. Fig. 6a–c shows the dilatometer curves (% expansion vs. temperature) of as-consolidated W20SiC pellets sintered at 1680 °C MA'd for 3 h, 6 h and 24 h, respectively. As can be seen in Fig. 6a–c, sintering behaviors of all pellets were similar. Furthermore, it was observed that the initial reducing effect of H₂ began approximately for all at the same temperatures (\sim 680 °C), even though its flow had started earlier. Moreover, this initial effect slowed down again at approximately the same temperatures (\sim 1000 °C), even though the H₂ flow was continuing. According to these curves, samples MA'd for 3 h, 6 h and 24 h have shrinkage values of 10.8%, 13.4% and 20.5%, respectively.

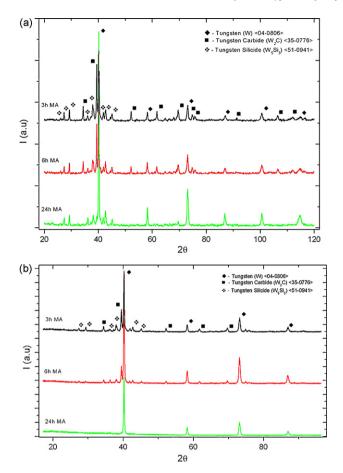


Fig. 7. (a) XRD patterns of W20SiC samples sintered at $1680 \, ^{\circ}$ C and MA'd for 3 h, 6 h and 24 h, (b) XRD patterns of W20SiC samples sintered at $1770 \, ^{\circ}$ C and MA'd for 3 h, 6 h and 24 h.

Sintering at $1680\,^{\circ}\text{C}$ was chosen as conventional sintering which is to heat the powder compact at a certain rate, followed by holding it at the highest temperature until the maximum density is reached. However, a second sintering regime is different which uses two steps in the heating schedule. The sample is first heated to a higher temperature ($1770\,^{\circ}\text{C}$) to achieve an intermediate density, then cooled down and held at a lower temperature ($1600\,^{\circ}\text{C}$). By means of the second sintering regime, the grain growth was tried to be inhibited. The characteristics of this regime are explained elsewhere [22,23]. Due to the fact that the dilatometer has a maximum operation temperature of $1700\,^{\circ}\text{C}$, second sintering regime was applied in LinnTM gas sintering furnace and sintering curves in Fig. 6a-c were extrapolated to $1770\,^{\circ}\text{C}$.

3.2.2. Microstructural evolution during sintering

Fig. 7a and b shows XRD patterns showing the effect of MA times taken from W20SiC samples sintered at $1680 \,^{\circ}$ C and $1770 \,^{\circ}$ C, respectively. As clearly seen in Fig. 7, phase compositions of the MA'd samples were changed as a result of the reactions during sintering process. W₂C phase is identified in the samples MA'd for 3 h and 6 h at $1680 \,^{\circ}$ C, which has a simple hexagonal Bravais lattice and $P\bar{3}ml$ space group with lattice parameters of $a = 0.299 \, \mathrm{nm}$ and $b = 0.473 \, \mathrm{nm}$ [24]. In addition, there are also some tungsten silicide phases (W₅Si₃ and WSi₂) identified, but it is somewhat difficult to identify the exact tungsten silicide phase since d-spacing values for these silicides are too close to each other. However, W₅Si₃ phase, which has a body-centered tetragonal Bravais lattice and I4/mcm space group with lattice parameters $a = 0.961 \, \mathrm{nm}$ and $b = 0.496 \, \mathrm{nm}$ [25] seems more likely to be stable in accordance

with the observation of Seng and Barnes [26] who expressed that W₅Si₃ is the most stable phase in the W-SiC system. Moreover, these results are consistent with the study of Baud et al. [27], where they detected W, W2C and W5Si3 in the SiC/W samples annealed above 1100 °C. Similarly, Son et al. reported the formation of W2C, W5Si3 and WC after hot-pressing W and SiC powders at 1700-1900 °C [28]. Also, Goesmann and Fetzer [29,30] showed that W₂C and W₅Si₃ were stable phases above 1200 °C and 1500 °C in W-SiC system, respectively. With increasing mechanical alloying, the intensities of W₂C diffraction patterns decrease and the intensities of W₅Si₃ peaks increase. For the W20SiC samples MA'd for 24 h and sintered at 1680 °C (Fig. 7a) only W and W₅Si₃ phases can be identified inferring that all the SiC was consumed to form W_5Si_3 and free carbon in the structure escaped out as some C_xH_y or CO_x gas. Weight loss of this sample (5%) was more than other samples, which supported this observation.

Thus, on the basis of Fig. 7a, it can be deduced that a chemical reaction between W_2C and SiC, and between SiC and W occurred at low sintering temperatures (1680 °C). Pertinent to this observation, it is appropriate here to list possible reactions, which Seng and Barnes [26] expressed as:

$$SiC + 8/3W_2C \rightarrow 1/3W_5Si_3 + 11/3WC$$
 (2)

$$SiC + 8/3W \rightarrow 1/3W_5Si_3 + WC \tag{3}$$

It is likely that the reaction product WC has reacted with W to form the W_2C present in the sintered W20SiC samples MA'd for 3 h and 6 h (Fig. 7a). On the other hand, W_2C is not detected for the sintered W20SiC sample MA'd for 24 h. The reason for this is not clearly understood.

Considering W20SiC samples sintered using second sintering regime (Fig. 7b), all possible phases (W, W_2C and W_5Si_3) are present in the sample MA'd for 3 h, but XRD peak intensities decrease in the sample MA'd for 6 h. In the sample MA'd for 24 h all the peaks except those of W disappeared. This is believed to be due to recrystallization triggered by severe plastic deformation induced during mechanical alloying. As a result, new strain free grains of W and SiC might have been formed.

SEM images of W20SiC samples sintered using second sintering regime are shown in Fig. 8a-c. SEM images comprise areas with varying contrasts. On the basis of EDS measurements taken from different areas, dull white areas consist of W and the darker, mottled white areas consist of W and W2C phases. In addition, grey areas consist of the silicide phases (Fig. 8a and b) and spherical shaped black areas are mainly pores. Whereas both W grains and W+W₂C regions have polygonal shapes, W₅Si₃ grains are finer but similar in shape. Further, comparing Fig. 8a-c with each other, it can be observed that MA has a definite effect on the evolution of the sintered microstructure resulting in the enlargement of all phases (W, W₂C, and W₅Si₃) with MA duration. Although the microstructures and shapes of the seen in Fig. 8a and b are similar, the size ranges of W and W_5Si_3 grains increase from 0.5 μm to 2 μm and 0.5 μm to $2.5 \, \mu m$ for the 3 h MA'd sample to 1 μm to 3 μm and 1 μm to 4 μm for the 6 h MA'd sample, respectively. Similarly, W+W2C regions are grown from a size range of 1-4 µm for the 3 h MA'd sample (Fig. 8a) to about 2–5 µm for the 6 h MA'd sample (Fig. 8b). Pores are mostly smaller than 0.5 µm in the W20SiC sample MA'd for 3 h (Fig. 8a) and are between 0.5 μm and 1 μm in size in the sample MA'd for 6 h (Fig. 8b). As seen in Fig. 8c, the microstructure of the sintered W20SiC sample MA'd for 24 h comprise only W grains and contain large voids and thus is different from those of the MA'd 3 h and MA'd 6h samples. The presence of only W phase in the 24h MA'd sample also confirms the XRD scan given in Fig. 7b.

Considering the W20SiC samples sintered at 1680 °C, the one MA'd for 3 h (Fig. 9a) consists of very small grains, but also contains a large number of pores and cracks due to the presence of

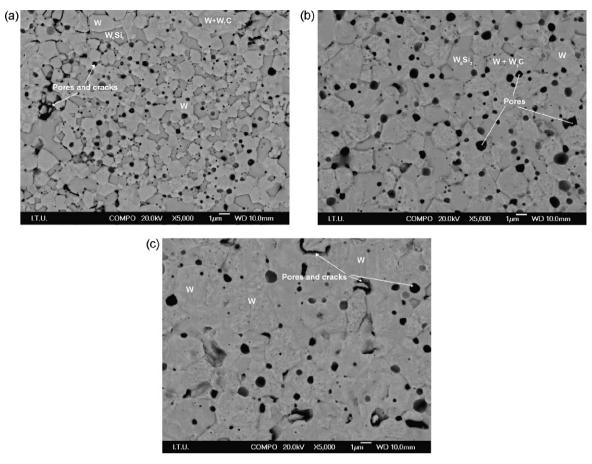
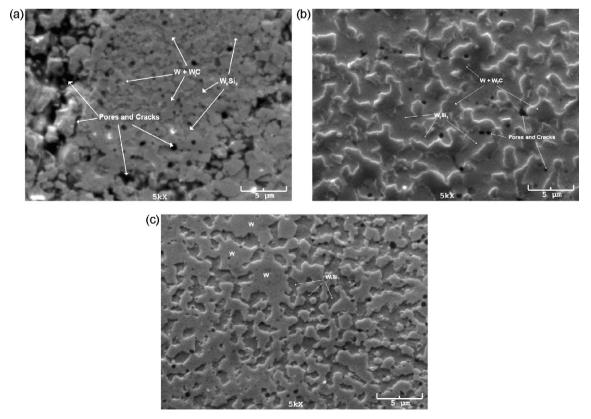


Fig. 8. (a), (b) and (c) SEM images of 3 h, 6 h and 24 h MA'd W20SiC samples sintered at $1770 \, ^{\circ}$ C.



 $\textbf{Fig. 9.} \ \, (a), (b) \ \, \text{and} \ \, (c) \ \, \text{SEM images of 3 h, 6 h and 24 h MA'd W20SiC samples sintered at 1680 } \\ \circ C, \ \, \text{respectively.}$

Table 2 Bulk density (g/cm 3) and relative density (%) values of W20SiC samples in the asconsolidated state and after sintering at 1680 $^{\circ}$ C and at 1770 $^{\circ}$ C.

	3 h MA'd	6 h MA'd	24 h MA'd
As-consolidated	8.67 (55.0%)	8.62 (54.0%)	8.01 (51.3%)
Sintered at 1680 °C	14.00 (89.0%)	14.60 (93.0%)	15.33 (97.7%)
Sintered at 1770 °C	14.88 (94.8%)	15.03 (95.8%)	14.94 (95.2%)

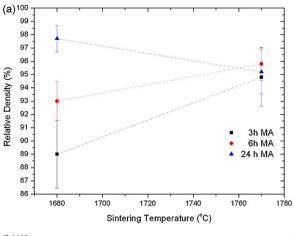
the brittle W_2C phase in substantial amounts. Moreover, the distribution of the pores and cracks is non-uniform, which makes the microstructure inhomogeneous. On the other hand, the sintered sample MA'd for 6 h (Fig. 9b) has a more uniform microstructure with larger grains and smaller amount of pores than the 3 h MA'd sample. It can be observed that although the microstructures of W20SiC samples MA'd for 24 h (Fig. 9c) and MA'd for 6 h (Fig. 9b) are similar, the former is finer than the latter. Moreover, as detected in Fig. 7a and observed in Fig. 9b and c, while the 6 h MA'd sample contains the W_2C phase in addition to the W and W_5Si_3 phases, 24 h MA'd sample has only two phases which are W and W_5Si_3 phases.

3.2.3. Density measurements

The results of the density measurements of both green compacts and sintered samples are given in Table 2. The theoretical density of W20SiC samples is calculated as 15.7 g/cm³. Relative densities of sintered samples changed with MA duration which varied from 89.0% to 97.7% for samples sintered at 1680 °C. On the other hand, those for W20SiC samples sintered using second sintering cycle are about 95%. According to these results, it is clear that the bulk density increases with increasing sintering temperatures for the samples MA'd for 3 h and 6 h as a result of thermal activation and increased atom diffusivity (Fig. 10a). Noting the fact that relative density values of 92% and above are accepted as high sintering densities [20,21], relative sinter densities of about 95% achieved in this study for all W20SiC samples MA'd at different times (3 h, 6 h and 24h) sintered using an upper and a lower sintering temperature (second sintering regime) are remarkable. Furthermore, sintering at 1680 °C yields an almost perfect bulk density value of 97.7% for the W20SiC sample MA'd for 24h which is the maximum sinter density value achieved in the present investigation. However, this is a result of a unique combination of sintering temperature and MA time. As seen in Table 2 and Fig. 10, bulk densities of the samples sintered at 1680 °C drastically increases with increasing MA duration, viz. 89.0%, 93.0% and 97.7% for 3 h, 6 h and 24 h, respectively. It is believed that two mechanisms contribute to the increase of bulk density with MA for the W20SiC samples sintered at 1680 °C. First of all, with increasing MA duration, mechanically activated sintering kinetics due to cold work stored energy enhances densification during sintering. Secondly, increasing MA duration causes more homogeneous distribution of SiC phase in the W-matrix which contributes to higher sinter densities. On the other hand, the fact that an almost constant sintering density (\sim 95%) value is achieved for all W20SiC samples sintered using the second sintering regime could be attributed to slower sintering kinetics [22,23,33-35] due to the "frozen" microstructure as reported in the literature [22,23,31–35].

3.2.4. Microhardness measurements

Since the samples produced in this research were very small, only microhardness measurements were used to investigate mechanical behavior of the samples. Vickers microhardness results of sintered W20SiC samples are given in Fig. 11 and are tabulated in Table 3. As seen in both Fig. 11 and Table 3, with increasing MA duration, hardness values decreased for both sintering temperatures. It is quite evident that hardness increases with increasing amount of W₂C in the W-matrix of the sintered microstructure. Thus, for all sintered W20SiC samples using both sintering conditions, decrease



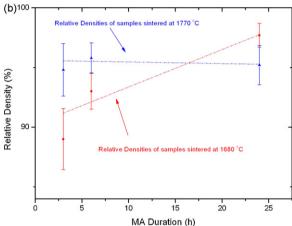


Fig. 10. (a) Relative density of W200SiC samples vs. sintering temperature and (b) relative density vs. amount of MA.

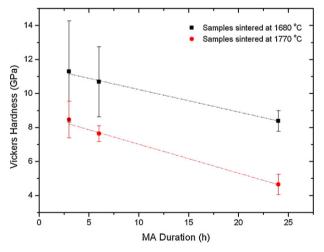


Fig. 11. Average microhardness values and standard deviations of W20SiC samples sintered at $1680\,^{\circ}\text{C}$ and $1770\,^{\circ}\text{C}$ as a function of MA time.

Table 3 Vickers microhardness results (GPa) of the W20SiC samples sintered at $1680\,^{\circ}$ C and at $1770\,^{\circ}$ C.

MA	3 h	6 h	24 h
Sintered at 1680°C Sintered at 1770°C	$\begin{array}{c} 11.30 \pm 2.98 \\ 8.47 \pm 1.08 \end{array}$	$10.69 \pm 2.06 \\ 7.64 \pm 0.47$	$\begin{array}{c} 8.39 \pm 0.61 \\ 4.65 \pm 0.60 \end{array}$

in hardness values is due to decrease in the amount of W₂C phase with increasing MA duration, as evidenced by XRD graphs given in Fig. 7. Considering the hardness values of the samples MA'd for 3 h and MA'd for 6 h sintered at 1680 °C, both have high hardness values (11.3 GPa for MA'd 3 h and 10.7 GPa for MA'd 6 h) and also have very high standard deviation values. From the SEM micrographs shown in Fig. 9, it can be observed that these high standard deviation values can be attributed to non-uniform nature of pore and crack distribution. High amount of W₂C phase is the reason for both high hardness values and the presence of cracks and pores in W20SiC MA'd for 3 h and 6 h followed by sintering at 1680 °C. As expected, the sample MA'd for 24 h and sintered at 1680 °C with no W2C phase has a lower microhardness value and a very low standard deviation as a result of its homogeneous microstructure. Hardness values of all W20SiC samples sintered using the second sintering regime were lower than those of sintered at 1680 °C, probably due to larger grain sizes achieved in the W20SiC samples sintered using the second sintering regime.

4. Conclusions

On the basis of the results of the present investigation, the following observations and conclusions can be drawn.

- 1. Only W phase with no other intermetallic phases has been identified in all MA'd W20SiC powders of the present investigation. Both average grain size and particle sizes of the MA'd powders decrease with MA duration. Average grain sizes, calculated on the basis of broadening of XRD peaks, have been predicted as 26 nm and 12 nm for W20SiC powders MA'd for 3 h and 6 h, respectively. In addition, low particle sizes like 202 nm and 195 nm are obtained for W20SiC samples after MA for 6 h and 24 h, respectively.
- 2. Using two different sintering regimes, W₂C and W₅Si₃ phases have been identified in the W-matrix for the sintered samples of W20SiC MA'd for 3 h and 6 h whereas W₂C is not present for the ones MA'd for 24 h. With increasing MA duration, the sintered microstructures become more homogeneous and pore-free.
- 3. Bulk densities of the samples sintered at 1680 °C varies between 89.0% and 97.7% with the maximum value achieved after MA for 24 h. On the other hand, constant relative sinter densities of about 95% are measured for all W20SiC samples MA'd at different times and sintered using an upper and a lower sintering temperature.
- 4. For all W20SiC composites sintered by two different sintering regimes, microhardness values decrease with increasing MA duration as a result of W₂C depletion with increasing MA duration. Of all sintered W20SiC samples, the one MA'd for 3 h and sintered at 1680 °C has the maximum microhardness value of about 11.3 GPa.

In overall, the present investigation has shown that mechanical alloying is very effective in preparing the W20SiC composites with high relative densities (\geq 95%) sintered at unconventionally low sintering temperatures (1680 °C and 1770 °C).

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