

Biotransformation of ruscogenins by *Cunninghamella blakesleeana* NRRL 1369 and neoruscogenin by endophytic fungus *Neosartorya hiratsukae*

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ARTICLE INFO

Article history:

Received 19 September 2017

Received in revised form

2 April 2018

Accepted 9 April 2018

Available online 22 April 2018

Keywords:

Microbial biotransformation

Ruscus

Ruscogenins

Neoruscogenin

Cunninghamella blakesleeana

Neosartorya hiratsukae

ABSTRACT

Biotransformation of steroidal ruscogenins (neoruscogenin and ruscogenin) was carried out with *Cunninghamella blakesleeana* NRRL 1369 and endophytic fungus *Neosartorya hiratsukae* yielding mainly P450 monooxygenase products together with a glycosylated compound. Fermentation of ruscogenins (75:25, neoruscogenin-ruscogenin mixture) with *C. blakesleeana* yielded 8 previously undescribed hydroxylated compounds. Furthermore, microbial transformation of neoruscogenin by endophytic fungus *N. hiratsukae* afforded three previously undescribed neoruscogenin derivatives. While hydroxylation at C-7, C-12, C-14, C-21 with further oxidation at C-1 and C-7 were observed with *C. blakesleeana*, *N. hiratsukae* biotransformation provided C-7 and C-12 hydroxylated compounds along with C-12 oxidized and C-1(O) glycosylated derivatives. The structures of the metabolites were elucidated by 1-D (¹H, ¹³C and DEPT135) and 2-D NMR (COSY, HMBC, HMQC, NOESY, ROESY) as well as HR-MS analyses.

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1. Introduction

The natural product drug discovery process involves isolation of new metabolites from natural sources, investigation of their biological activities and semi-synthesis of more active analogs. Microbial transformation plays a vital role in the preparation of new oxygenated derivatives due to the stereo- and regio-selective reaction capability (Muffler et al., 2011; Swizdor et al., 2012; Wang et al., 2015). Recently, the microbial biotransformation has been used for numerous purposes: i) to overcome the problems encountered in synthesis reactions; ii) to establish biosynthetic pathways; iii) to increase/decrease the bioactivity/toxicity profiles of the lead molecules; iv) to assist structure-activity relationship studies (Chen et al., 2010a). In addition to these, it is a useful method to model mammalian drug metabolism (Piska et al., 2016).

Metabolism of drugs is vital for the safety and efficacy, and the microbial models are advantageous due to their competences

enabling researchers to identify minor metabolites. It has been proven, that cytochrome P450 monooxygenase enzymes that exist in all eukaryotic microorganisms, catalyze the hydroxylation of steroidal compounds, especially *Cunninghamella* species (Asha and Vidayavathi, 2009; Nelson, 1999).

In recent years, endophytic organisms attract great deal of attention. They are bacterial or fungal microorganisms that colonize living internal tissues of plants without causing any disease symptoms, and they constitute one of the most exciting groups of microorganisms. While producing novel bioactive compounds, they can also interact with their hosts via special enzymes and chemicals, which would make them useful sources as biocatalysts for the biotransformation of natural products (Kumar and Ahmad, 2013; Suryanarayanan et al., 2012).

Ruscus aculeatus (Butcher's broom, Rusci rhizome, Asparagaceae) have been used traditionally against chronic venous disorders. Its extracts have strong anti-inflammatory activity, act as anti-elastase, and decrease capillary permeability (Bouskela et al., 1994; Facino et al., 1995). The major steroidal sapogenins, neoruscogenin (NR) and ruscogenin (R), namely ruscogenins as a mixture of 75:25 (NR:R), are the secondary metabolites responsible

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for aforementioned bioactivities, and they are used in clinics to treat vascular disorders such as varicose and hemorrhoid. In various pharmaceutical preparations, *Ruscus* dried root powders or its hydroalcoholic extracts are used in combination with phenolic compounds like trimethylhesperidin chalcone, ascorbic acid or *Melilotus officinalis* extract for vascular health (Boyle et al., 2003; Jawien et al., 2017). The European Pharmacopoeia prescribes not less than 1.0% of total saponinins expressed as ruscogenins, and European Scientific Cooperative on Phytotherapy recommends oral use of 7–11 mg/day. However, there are still unknown matters in regards to the absorption of the active substances after oral administration, and whether they undergo hepatic metabolism impairing their activity at lymphatic level (EMA, 2008; ESCOP, 2003).

Biotransformation and semi-synthesis studies performed on ruscogenins have so far been limited and there has been no report on neoruscogenin (Chen et al., 2010a, 2010b). The aim of this study was to estimate metabolism of ruscogenins and obtain new analogs deriving from neoruscogenin. Herein, we report isolation and structural elucidation of 11 previously undescribed biotransformation products of ruscogenins with diverse chemistry (Figs. 1 and 2).

2. Results and discussion

The literature survey prompted us to use *C. blakesleeana* as biocatalyst to predict the possible metabolism products of the major bioactive constituents of *Ruscus aculeatus*. Furthermore, in order to reveal the potential use of endophytes as biocatalyst, *N. hiratsukae* was used to transform neoruscogenin. After 10 days of incubation period, a number of compounds (**1–10**) were purified from the microbial broths, and their structures were established by 1D, 2D NMR and HRMS experiments. The ^1H and ^{13}C NMR spectral data of these compounds are presented in Tables 1–3.

2.1. Metabolite 1

In the HR-MS of **1**, the major ion peak was observed at m/z 461.2899 $[\text{M}+\text{H}]^+$ (calcd. 461.2903 for $\text{C}_{27}\text{H}_{41}\text{O}_6$), indicating 8 degrees of unsaturation as in NR as well as the presence of two

hydroxyl groups due to 32 amu difference. Inspection of the ^1H NMR spectrum revealed additional signal at δ 4.94, whereas two new low-field carbon resonances appeared at δ 67.3 and 86.8 in the ^{13}C NMR spectrum, which were attributed to secondary and tertiary alcohol groups based on the HSQC data. Detailed examination of the COSY spectrum led to the deduction of 4 major spin systems (SS): SS1 \rightarrow H-1(O)/H₂-2/H-3(O)/H₂-4 (A ring); SS2 \rightarrow H-6 (olefinic)/H-7(O)/H-8/H-9/H₂-11/H₂-12 (B and C rings); SS3 \rightarrow H₂-15/H-16(O)/H-17/H-20/H₃-21; SS4 \rightarrow H₂-23/H₂-24 (allylic coupling)/H₂-27(allylic coupling)/H₂-26(O) (Fig. 3). The SS2 correlation between H-6 (δ 5.97) and the new resonance appeared at δ 4.94 verified that the first hydroxylation was at C-7. The carbon signal at δ 86.8 was readily assigned to C-14 on the basis of a long-distance correlation with the singlet methyl proton ascribed for H₃-18 (δ 1.20) and H₂-15 (δ 2.47 and 3.18) (Fig. 3). Moreover, separation of the SS2 and SS3 due to the presence of a quaternary carbon (C-14) verified the position of the second hydroxyl group.

In the 2D-NOESY spectrum, the correlation of alpha oriented H-9 (δ 2.45) with H-7 (δ 4.94), and H-7's correlation with the exchangeable proton of C-14(OH) (δ 5.40) in turn, revealed the relative configuration of C-7(OH) and C-14(OH) as β and α , respectively (Fig. 4). Consequently, the structure of **1** was established as 7 β ,14 α -dihydroxy neoruscogenin.

2.2. Metabolite 2

The major ion peak at m/z 497.2531 $[\text{M}+\text{Na}]^+$ (calcd. 497.2515 for $\text{C}_{27}\text{H}_{38}\text{O}_7\text{Na}$) in the HR-MS spectrum of metabolite **2** displayed 46 amu increase compared to NR. In the ^1H NMR spectrum of metabolite **2**, the observed resonances were mostly identical with neoruscogenin, except a low-field signal at δ 4.80 suggesting a hydroxylation. In the ^{13}C NMR spectrum, the chemical shift differences were more noticeable with a carbonyl signal at 201.0 ppm, and a pair of tri-substituted olefinic system signals (δ 128.4 and 165.6; deduced by HSQC and HMBC). Two additional carbon resonances in the low-field were also noticed (δ 71.6 and 86.9). The significant down-field shift for one of the aforementioned double bond resonances (δ 165.6) suggested an alpha-beta unsaturated ketone functionality in the structure. The absence of characteristic H-6 (δ 6.01) and H-7 correlation in the COSY spectrum implied a

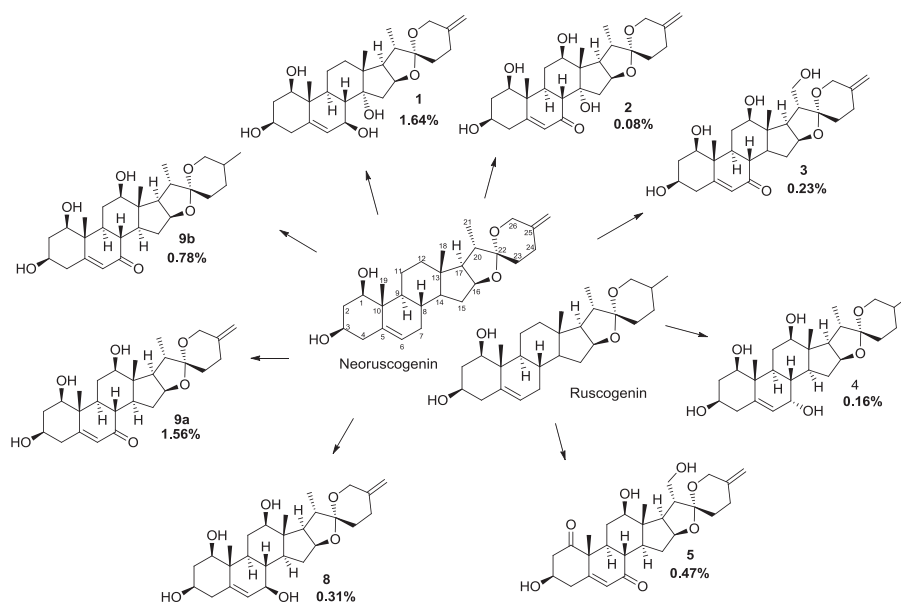


Fig. 1. Biotransformation products of ruscogenins by *Cunninghamella blakesleeana*.

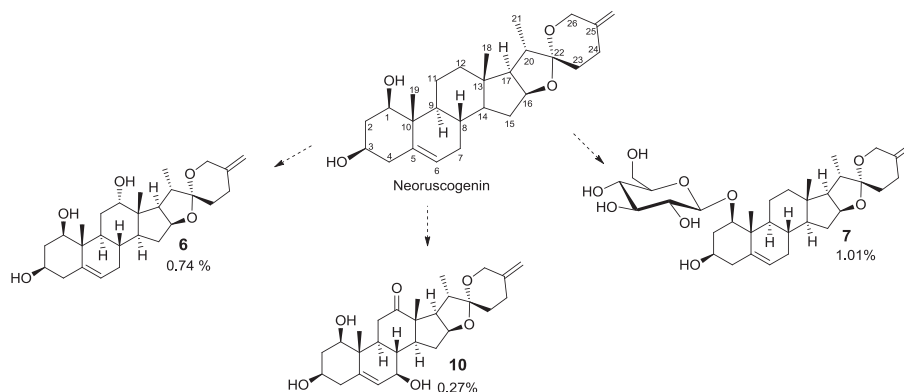


Fig. 2. Biotransformation products of neuruscoegenin by *Neosartorya hiratsukae*.

Table 1
¹H-NMR data of metabolites 1–7.

Position	1	2	3	4	5	6	7
	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)	δ_{H} (ppm), J (Hz)
1	3.82 m	3.82 dd (4/11.5)	3.79m	3.81 m	–	3.79 m	3.92 m
2	2.23, 2.59 m	2.27 m, 2.64 m	2.27 m, 2.65 m	2.73 m	2.94 m, 3.03 m	2.21 m, 2.62 m	2.12 m, 2.71m
3	3.92 m	3.89 m	3.94 m	3.97 m	4.23 m	3.94 m	3.83 m
4	2.67 m	2.73 d (8)	2.74 m	2.69 m	2.94 m	2.68 m	2.56 m
5							
6	5.97 bs	6.01 s	6.02 s	6.07 d (3.6)	6.03 s	5.59 d (6.5)	5.55 d (4.6)
7	4.94 m			4.14 m		1.95 m	1.84 m
8	2.32 dd (8.8, 12.0)	2.97 m	2.61 m	1.79 m	2.59 dd (11.5/12.0/13.0)	2.22 m	1.54 m
9	2.45 m	2.97 m	1.96 m	2.35 m	2.37 m	1.55 m	1.51 m
10							
11	1.96 ddd (4.0/4.4/12.8/13.2), 2.98 d (12.8)	2.26 m, 3.34 m	2.05 m, 3.25 m	2.13 d (9.2), 3.35 dt (3.2/7.2)	1.82 m, 2.38 m	2.03 m, 3.30 td (5.0/5.5/6.0/17)	2.88 m, 1.38 m
12	1.56 m, 2.41 m	4.80 m	3.73 m	3.79 m	3.74 dd (3.0/3.5/11.0)	3.68 m	1.36 m, 1.65 m
13							
14			1.71 m	2.17 m	1.67 m	1.24 m	1.13 m
15	3.18 dd (6.4/7.2/7.2), 2.47 m	2.40 ddd (6.0/7.0/7.5) 3.60 ddd (6.0/7.5/7.5)	1.90 m, 3.38 m	1.71 m 2.65 m	1.92 m 3.37 dddd (6.0/7.0/7.5)	1.63 m, 1.96 m	1.42 m, 1.98 m
16	5.17 ddd (7.2/7.2/7.6)	5.21 ddd (7/7.5)	4.79 m	4.69 ddd (7.5/7.5/8.0)	4.77 m	4.59 ddd (9.0/9.0/10.5)	4.43 m
17	2.81 dd (6.8/7.6)	3.14 m	2.42 m	2.28 m	2.40 m	2.18 m	1.76 m
18	1.20 s	1.34 s	1.19 s	1.24 s	1.17 s	1.15 s	0.86 s
19	1.39 s	1.48 s	1.44 s	1.36 s	1.46 s	1.34 s	1.24 s
20	2.12 dd (6.8/6.8)	2.35 m	2.53 m	2.25 m	2.53 m	2.22 m	1.91 m
21	1.11 d (6.8)	1.44 d (7)	4.12 m	1.43 d (6.4)	4.04 m	1.38 d (6)	1.03 d (6.5)
22							
23	1.83 m	1.83 m	1.87 m	1.81 m	1.86 m	1.81 m	1.85 m
24	2.25 m, 2.73 m	2.26 m, 2.73 d (8)	2.22 m, 2.66 m	1.72 m	2.19 m, 2.66 m	2.25 m, 2.73 m	2.23 m, 2.67 m
25				1.56 m			
26	3.96 d (12), 4.42 d (12)	4.00 d (12.5), 4.45 d (7.0)	3.95 d (12), 4.42 d (12)	3.54 m, 3.69 m	3.93 d (12), 4.41 d (12.5)	4.04 d (12), 4.49 d(12)	4.03 d (12), 4.44 d (12)
27	4.75 d (4.8)	4.77 d (10.5)	4.75 s, 4.79 s	0.65 d (4.4)	4.74, 4.77 s	4.77 s, 4.80 s	4.79 s, 4.80 s
1'							4.95 d (8)
2'							4.03 dd (7.5/9.5)
3'							4.23 t (7.5/9)
4'							4.13 t (9/9.5)
5'							3.93 m
6'							4.33 m, 4.57 dd (3/11.5)

conjugated system formed on the ring B (–C5=C6–C7(=O)–). Based on the DEPT 135 and HSQC spectra, it was inferred that the δ 86.9 signal was a quaternary carbon, whereas the 71.6 resonance was a methine correlating with a proton at δ 4.80. The planar structure of **2** was deduced by detailed inspection of the HMBC spectrum. The $^3J_{\text{H-C}}$ correlations of H₃–18 (δ 1.34) with C–17 (δ 58.2), C–13 (δ 50.8) and two oxygenated carbon signals at δ 71.6 and 86.9, helped us to

locate the hydroxyl groups at C–12 and C–14, respectively. In the HMBC spectrum, the ketone signal at δ 201.0 displayed a correlation with δ 2.97 resonance. In the HSQC spectrum, δ 2.97 proton corresponded with a carbon signal at δ 45.2, which was attributed to C–9 based on its long-range correlation with H₃–19 (δ 1.48). Thus, as suggested above, the carbonyl group was located at C–7. Inspecting 2D-NOESY spectrum and comparing **2**'s carbon NMR data with

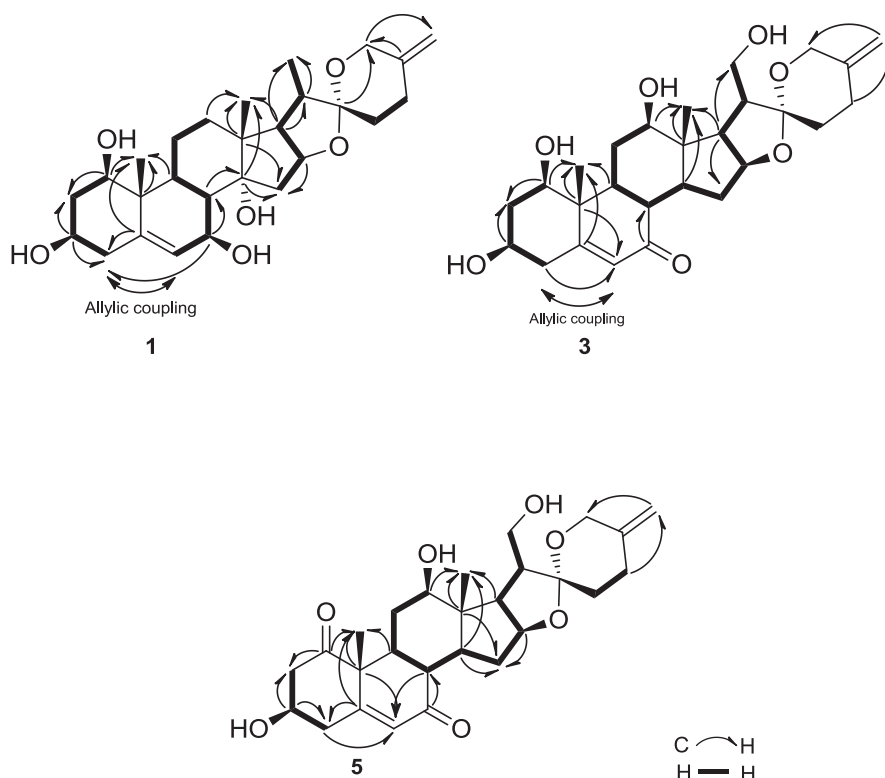


Fig. 3. COSY and key HMBC correlations of metabolites **1**, **3** and **5**.

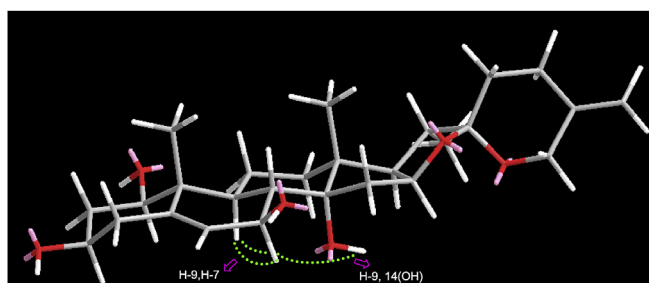


Fig. 4. Minimized energy conformer of metabolite **1** and key nOe correlations.

those of established metabolites assisted us to deduce relative configurations of the oxygenated positions. The absence of a correlation between the β -oriented H₃-18 (in **3** 1.34) and H-12 (in **3** 4.80) indicated that C-12(OH) was β -oriented, whereas the identical C-14 data of metabolites **1** and **2** (in **3** 86.8 and 86.9, respectively) revealed an alpha configuration for C-14(OH). Consequently metabolite **2** was elucidated as 12 β ,14 α -dihydroxy neuroscogenin-7-one.

2.3. Metabolite **3**

The molecular formula was found to be C₂₇H₃₈O₇ based on the major ion peak at m/z 497.2509 [M+Na]⁺ (calcd. 497.2515 for C₂₇H₃₈O₇Na). Initial inspection of the ¹H NMR spectrum of **3** revealed the absence of characteristic secondary CH₃-21 resonance as well as appearance of two down-field signals at δ 4.12 (2H) and 3.73 (1H). The HSQC spectrum showed a ¹J_{C-H} correlation between the δ 4.12 signal and a methylene-type carbon at δ 63.2, suggesting hydroxylation of C-21 to afford an oxymethylene group extending from C-20. In the ¹³C NMR spectrum of **3**, the identical resonances

(δ 201.4, 164.4 and 127.6) in comparison to **2** suggested an α,β -unsaturated carbonyl system of C-5, C-6 and C-7, which was further confirmed by H-6's (δ 6.02) characteristic down-field shift (ca. 0.4 ppm). The new oxymethine signal at 78.4 ppm showed ³J_{H-C} long-distance correlation with the singlet methyl proton of H₃-18 (δ 1.19) in the HMBC spectrum, assisting us to locate the other hydroxyl group at C-12 (Fig. 3). Moreover, the hydroxyl group at C-21 was also confirmed with the key HMBC's between C-17, C-20 and H₂-21, and C-21 to H-17. In the NOESY spectrum, H-12 (δ 3.73) displayed correlations with alpha-oriented H-14 (δ 1.71) and H-17 (δ 2.42), establishing β -configuration of C-12(OH). In conclusion, the structure of metabolite **3** was determined to be 12 β ,21-dihydroxy neuroscogenin-7-one.

2.4. Metabolite **4**

In the ¹H NMR spectrum of metabolite **4**; disappearance of characteristic H-26 and H-27 signals of NR in the low-field, and observation of a new methyl signal in the down-field region (δ 0.65) implied that the compound was a ruscogenin derivative. Correspondingly, loss of the characteristic olefinic signals (C-25 and C-27) of neuroscogenin with an additional methyl resonance at 17.1 ppm in the ¹³C NMR spectrum, and the major ion peak at m/z 463.3059 [M+H]⁺ (463.3060 calcd.) in the HR-MS spectrum substantiated our first impression of **4** being a ruscogenin metabolite. The mass data exhibiting 32 amu increase over ruscogenin, and appearance of two additional carbon signals at δ 79.1, 64.2 suggested hydroxylations. In the HMBC spectrum, the ³J_{H-C} correlations of characteristic H₃-18 resonance (δ 1.24) with C-13 (45.5), C-14 (49.0), C-17 (62.9) and the new signal at δ 79.1 (C-12) verified the position of first hydroxylation at C-12. In the COSY spectrum, a strong correlation from olefinic signal at δ 6.07 (H-6) with the new resonance at δ 4.14 (H-7) helped us to locate the second hydroxyl

group at C-7. To determine the relative stereochemistry of **4**, 2D-NOESY and ROESY spectra were inspected. The correlation of H-12 with alpha oriented H-9 (δ 2.35), and a cross peak between H-7 and beta oriented H-8 (δ 1.79) revealed the configurations of OH groups. As a result, metabolite **4** was established as 7 α ,12 β -dihydroxy ruscogenin.

2.5. Metabolite 5

The molecular formula of **5** was determined to be C₂₇H₃₆O₇ based on the major ion peak at m/z 495.2363 [M+Na]⁺. In the ¹H NMR spectrum of **5**, together with characteristic NR signals, three additional low-field proton signals (δ 4.23, 4.04, 3.74) were observed. The disappearance of distinctive secondary methyl group (C-21) was also noticed readily as in metabolite **3**. Besides, ¹³C NMR and DEPT135 revealed extra signals in the low-field: two carbonyl resonances (δ 208.2, 199.7), an oxymethine (δ 77.8) and an oxymethylene (δ 63.2). On the basis of these data, it was evident that, C-21 and an additional position were hydroxylated on the NR framework. Ketone formation at C-7 position was clear due to deshielding of H-6 (δ 6.03) and C-5 (δ 159.8). However, C-5's *ca.* 5 ppm up-field shift compared to those of **2** and **3** implied an alteration on the ring A. In the HMBC spectrum (Fig. 3), strong ³J_{H-C} correlations of H₃-19 (δ 1.46) with δ 208.2, C-5 (δ 159.8), C-9 (δ 42.1) and C-10 (δ 54.2) signals suggested oxidation of the secondary alcohol at C-1, whereas the correlation of H₃-18 (δ 1.17) with the new oxygenated methine carbon at δ 77.8 implied the second hydroxylation position as C-12. All of these deductions were also verified by establishing the spin systems of metabolite **5** via examination of the COSY, TOCSY and HSQC spectra: SS1 → H₂-2/H-3(O)/H₂-4 (A ring); SS2 → H-12(O)/H-11/H-9/H-8/H-14/H₂-15/H-16(O)/H-17/H-20/H₂-21; SS3 → H₂-23/H₂-24 SS4 → H₂-27 (allylic coupling)/H₂-26(O). The stereochemistry of the hydroxyl group at C-12 was determined by examining the NOESY and ROESY spectra, which exhibited a cross peak from alpha oriented H-14 (δ 1.67) to H-12 (δ 3.74) confirming the orientation of C-12(OH) as beta. On the basis of these evidence, the structure of compound **5** was elucidated as 12 β ,21-dihydroxy neoruscogenin-1,7-dione.

Metabolites **6** and **7** were obtained from the biotransformation study of *N. hiratsukae*.

2.6. Metabolite 6

The HR-MS spectrum of **6** provided a major ion peak at m/z 445.2949, indicating 16 amu increase over NR. In the ¹H NMR spectrum of **6**, in addition to the characteristic low-field signals of neoruscogenin, an additional resonance at δ 3.68 was observed (Table 2). In the ¹³C NMR and DEPT135 spectra, the additional oxymethine signal at δ 79.3 undoubtedly demonstrated monohydroxylation on the skeleton, while a key HMBC between the new oxymethine carbon signal and H₃-18 methyl group (δ 1.15) helped us locating the hydroxyl group at C-12. The relative configuration at the new stereocenter was established based on the NOESY data, displaying correlations between H-12 (δ 3.68) and β -oriented H₃-18/H-20 protons (δ 1.15 and 2.22, respectively). Consequently, the structure of **6** was determined to be 12 α -hydroxy neoruscogenin.

2.7. Metabolite 7

In the HR-MS of **7**, the major ion peak was observed at m/z 613.3353 [M+Na]⁺ (calcd. 613.3353 for C₃₃H₅₀O₉Na), which was suggesting glycosidation of the neoruscogenin skeleton due to 162 amu difference (C₆H₁₀O₅). Inspection of the ¹H NMR spectrum revealed an anomeric proton at δ 4.95 (d, J = 8.0 Hz), whereas the additional resonances in the ¹³C NMR spectrum deriving from a

hexose unit were consistent with the presence of a β -glucopyranose moiety. Comparing the spectral data of **7** with neoruscogenin and the other metabolites, the significant down-field shift of C-1 (δ 82.9, Δ_{ppm} = 6–8) helped us locating the glucopyranose unit. Additionally, the ³J_{H-C} correlation of C-1 (δ 82.9) with the anomeric proton signal (δ 4.95) in the HMBC spectrum verified the glycosidation site as C-1. On the basis of these results, the structure of **7** was established as neoruscogenin-1-O- β -glucopyranoside.

2.8. Metabolites 8, 9a, 9b and 10

Metabolites **8–10** were not brought to a good state of purity due to purification difficulties. However, the structures of these metabolites were identified based on detailed inspection of 1- and 2D NMR spectra.

The major ion peak at m/z 483.2717 [M+Na]⁺ in the HR-MS spectrum of metabolite **8** displayed 32 amu increase over NR, suggesting a dihydroxy analog. In the low-field region of the ¹H NMR spectrum, two additional low-field protons (δ 3.27 and 3.65) were observed (Table 3). Besides, in the ¹³C NMR spectrum, two extra down-field carbon signals at δ 72.7 and 80.3 were also noted. In order to deduce oxidation positions, the 2D NMR spectra were inspected in detail. Full assignments of the proton and carbon signals of **8** secured by the COSY and HSQC spectra implied the position of hydroxylations as C-7 and C-12. In the HMBC spectrum, the long-range correlations of the olefinic carbon signals at 141.4 and 130.9 ppm (C-5 and C-6, respectively) with H-7 (δ 3.65), and C-12 (δ 80.3) and H₃-18 (δ 0.81) verified the proposed transformations. For establishing stereochemistry of **8**, a 2D-NOESY experiment was performed. Alpha oriented H-9 (δ 1.35) exhibited correlations with H-7 (δ 3.65) and H-12 (δ 3.27), while H-17 α (δ 1.87) correlated with H-12, substantiating co-facial orientation of C-7 and C-12 OH groups as β . Based on these findings, the structure of **8** was determined as 7 β ,12 β -dihydroxy neoruscogenin. In the ¹H NMR spectrum of **9**, appearance of two distinctive H-16 signals and presence of characteristic methyl resonances for both starting frameworks (R and NR) point to a mixture of two metabolites, NR being the major one (**9a**). In the ¹³C NMR spectrum, similar with metabolite **4**, a weak methyl resonance at 16.2 ppm was attributed to the C-27 of the minor ruscogenin metabolite (**9b**). These suggestions were also substantiated with a major peak at m/z 459.2732 [M+H]⁺ and a minor peak at 461.3885 [M+H]⁺ in the HR-MS spectrum, indicating 30 amu difference compared to NR and R, respectively. All the proton and carbon resonances of metabolite **9a** were secured by COSY, HSQC and HMBC spectra, verifying C-12 hydroxylation as in **2–5** and **8**, and oxidation of C-7 to give ketone functionality as in compounds **2**, **3** and **5**. In the case of stereochemical assignment of C-12(OH), the chemical shift and coupling constant values obtained for metabolite **1** and **9a** were taken into consideration, revealing same configuration for both compounds as beta for the hydroxyl group. This suggestion was also substantiated with the nOe correlation of H-12 (δ 3.25) and alpha oriented H-9 (δ 1.79), H-14 (δ 1.41) and H-17 (δ 1.84) resonances in the 2D-NOESY spectrum. When the 1D and 2D-NMR spectra were also inspected for the minor constituent **9b**, completely identical structural conclusions were achieved for ruscogenin skeleton. Consequently, metabolites **9a** and **9b** were elucidated as 12 β -hydroxy neoruscogenin-7-one and 12 β -hydroxy ruscogenin-7-one, respectively. The HR-MS spectrum of **10** provided a major peak at m/z 481.2566, [M+Na]⁺ (calcd. 481.2566), supporting a molecular formula of C₂₇H₃₈O₆Na. In the ¹H NMR spectrum, in addition to typical neoruscogenin signals, the appearance of a proton resonance at δ 4.09, which was correlating with δ 70.6 resonance in the HSQC spectrum, implied transformation of a methylene carbon to oxymethine carbon via hydroxylation in the steroid framework. This

Table 2
¹³C NMR data of metabolites 1–7.

Position	1	2	3	4	5	6	7
	δ_c (ppm)	δ_H (ppm)	δ_H (ppm)	δ_H (ppm)	δ_H (ppm)	δ_c (ppm)	δ_c (ppm)
1	78.3	76.2	76.3	77.4	208.2	77.8	82.9
2	43.9	42.9	42.7	43.4	48.2	43.2	37.3
3	68.2	66.1	66.3	67.3	66.6	67.9	67.8
4	43.9	43.3	43.4	44.2	41.5	43.6	43.2
5	140.7	165.6	164.4	142.9	159.8	140.1	139.2
6	132.1	128.4	127.6	128.1	127.5	124.2	124.7
7	67.3	201.0	201.4	64.2	199.7	31.9	31.7
8	45.7	48.9	44.7	37.7	43.9	42.7	32.8
9	43.9	45.2	49.6	42.7	42.1	50.6	50.1
10	44.1	45.2	45.2	43.5	54.2	43.5	42.3
11	24.0	33.6	30.6	34.1	30.7	34.2	23.8
12	33.1	71.6	78.4	79.1	77.8	79.3	40.0
13	46.0	50.8	46.4	45.5	47.1	45.7	40.0
14	86.8	86.9	48.8	49.0	49.0	55.2	55.8
15	43.4	42.1	32.2	31.7	33.6	31.9	32.1
16	83.1	82.5	82.6	81.1	82.4	81.3	81.2
17	59.9	58.2	59.4	62.9	59.4	62.8	62.8
18	20.7	14.0	11.0	11.3	10.9	11.1	10.9
19	13.9	12.0	11.7	12.6	16.8	13.7	14.0
20	42.4	42.9	50.2	42.7	59.6	42.7	41.7
21	15.7	14.5	63.2	14.2	63.2	14.1	14.6
22	110.1	109.9	108.9	109.3	109.2	109.5	109.4
23	33.8	33.2	33.8	30.4	33.6	33.1	31.8
24	29.4	28.8	29.4	31.7	28.6	28.8	28.9
25	144.9	144.3	142.2	29.0	143.8	144.3	144.2
26	65.3	64.7	63.1	73.8	64.5	64.8	63.4
27	108.9	108.5	109.0	17.1	108.8	108.4	108.5
1'							101.2
2'							75.2
3'							78.3
4'							72.1
5'							77.9
6'							63.4

new low-field proton was ascribed to H-7 on the basis of its correlation to characteristic olefinic signal of H-6 (δ 5.93) in the COSY spectrum. As well, a carbonyl signal at δ 213.3, which was lacking in the ¹³C NMR spectrum due to its low intensity, displayed a strong long-range correlation with the methyl resonance at δ 1.20. The latter was assigned to H₃-18 because of its long-range correlation to C-17, which also exhibited ³J_{H-C} correlation to H₃-21. As a result of these key HMBC's, the position of the carbonyl group was secured as C-12. The beta oriented hydroxyl group at C-7 was readily deduced via the correlation of alpha oriented H-14 (δ 1.92) with H-7 (δ 4.09) in the NOESY spectrum. Thus, the structure of **10** was determined as 7 β -hydroxy neoruscogenin 12-one.

3. Conclusions

Recent studies performed on steroids showed that *Cunninghamella* species have high catalytic activity (Baydoun et al., 2017; Xiao et al., 2011; Xu et al., 2015), which can also be used for investigating the metabolism of drugs because of their abilities of catalyzing Phase I and II reactions similar to the mammalian enzyme systems. In continuation of our biotransformation studies performed on ruscogenins by *C. blakesleeana* (Özçınar et al., 2016), a number of previously unreported compounds were obtained. The compounds were mainly hydroxylation products including C-7, C-12, C-14 and C-21 positions, whereas further oxidations at C-1 and C-7 were encountered.

All the metabolites were C-7 hydroxylated or oxidized analogs as parallel to other studies performed on steroidal compounds (Ahmad et al., 2017; Kozłowska et al., 2017; Xu et al., 2015). This modification, catalyzed by fungal CYP450 monooxygenase and dehydrogenase enzymes, readily takes place due to the allylic

nature of C-7 position (Ortiz de Montellano and Nelson, 2011). Additionally, another major hydroxylation location was C-12, except compound **1**. Interestingly, this transformation was stereoselective [12- β (OH)] compared to the regioselective C-7 hydroxylation. The monooxygenation reactions occurred at the tertiary C-14 (**1** and **2**) and C-21 methyl group (**3** and **5**) have been encountered rarely on steroidal framework, as non-reactive centers, they are challenging positions for modification via chemical synthesis (Faramarzi et al., 2008; Kolet et al., 2014; Lamm et al., 2007; Wu et al., 2011).

In order to exploit the catalytic capacity of the endophytic fungi to obtain new neoruscogenin derivatives, *N. hiratsukae* was utilized, providing CYP450 monooxygenase products mainly as in *C. blakesleeana* study. Moreover, observation of further oxidation at C-12 to yield ketone functionality (**10**) and the phase II metabolite **7** as glycosylated derivative were noteworthy.

Further studies are in progress with different endophytic fungi to include more compounds to the metabolite library, as well bioactivity studies are being pursued for the compounds with sufficient quantity.

4. Experimental

4.1. General experimental procedures

Mass spectrometry analysis was performed on an Agilent 1200/6530 Instrument –HRTOFMS. The 1D and 2D NMR spectra were obtained on Varian Oxford AS400 and Bruker DRX-500 instruments. Proton and carbon chemical shifts are relative to the solvent signals. Column chromatography experiments were carried out on silica gel 60 (40–63 μ -Merck), Sephadex LH-20 (GE

Table 3
¹H and ¹³C NMR data of metabolites **8**–**10**.

Position	8		9a(9b)		10	
	δ_{H} (ppm), J (Hz)	δ_{C} (ppm)	δ_{H} (ppm), J (Hz)	δ_{C} (ppm)	δ_{H} (ppm), J (Hz)	δ_{C} (ppm)
1	3.38 m	78.5	3.49 dd (4.5/12)	75.8	3.70 m	76.8
2	1.61 m, 1.97 m	42.5	1.71 m, 2.10 m	40.8	2.21 m, 2.60 m	42.8
3	3.44 m	68.6	3.57 m	65.8	3.94 m	67.2
4	2.23 m	42.8	2.40 dd (11.5/13.0)	41.8	2.65 m	42.9
5		141.4		164.7		104.2
6	5.42 s	130.9	5.80 s	126.9	5.93 s	130.7
7	3.65 d (10.5)	72.7		202.6	4.09 m	70.6
8	1.49 m	40.8	2.45 m	44.1	2.23 m	40.4
9	1.35 ddd (3.6/4.4/7.6)	48.3	1.79 m	49.4	1.94 m	42.3
10		43.8		44.8		50.5
11	1.54 m, 2.51 m	34.2	2.27 m, 2.55 m	32.6	2.95 m, 3.66 d (5)	40.3
12	3.27 dd (4.0/4.4/6.4)	80.3	3.25 m	77.9		213.3
13		47.0		47.7		55.5
14	1.22 m	55.7	1.41 m	29.8	1.92 m	42.3
15	1.66 m, 2.35 m	35.0	1.50 m, 2.84 m	32.6	2.25 m, 2.95 m	34.5
16	4.45 ddd, (7.2/7.6/7.2)	82.8	4.51 ddd (6.5/7.5/8.0/8.5)	81.2	4.55 ddd (7.0/7.0/8.5)	80.6
17	1.87 m	62.7	1.84 m	60.8	2.85 m	53.5
18	0.81 s	11.3	0.80 s	9.7	1.20 s	15.8
19	1.10 s	13.5	1.29 s	10.6	1.31 s	12.7
20	1.91 m	43.7	1.91 m	42.3	1.94 m	42.3
21	1.02 d (6.1)	14.0	1.04 d (7.5)	12.8	1.29 d	14.0
22		110.8		109.6		108.5
23	1.74 m	34.1	1.73 m	32.6	1.78 m	33.3
24	2.26 m, 2.57 m	29.5	2.27 m, 2.55 m [1.71 m]	28.1	2.24 m, 2.68 m	29.4
25		145.3	[1.62 m]	143.7 [28.5]		143.9
26	3.83 d (11.8), 4.28 d (12.5)	65.9	3.85 d (12), 4.30 d (12) [3.37 d 11.0] [3.48 dd 4.5/11.7]	64.4 [66.5]	4.01 d (13), 4.43 d (12)	64.6
27	4.73 s, 4.77 s	108.9	4.76 s, 4.79 s [0.81 d 4.1]	107.8 [16.1]	4.77 s, 4.82 s	109.1

*Compounds **9a** and **9b** were isolated as a mixture for which most of the ¹H NMR data were superimposable. The most distinguishing ¹H NMR data of the minor metabolite [**9b**] are given above.

Healthcare) and RP (C-18, 25–40 μm) (Merck). TLC analyses were carried out on Silica gel 60 F254 (Merck) and RP-18 F254s (Merck) plates. Compounds were detected by UV and 20% H₂SO₄/water spraying reagent followed by heating at 105 °C for 1–2 min. Optic rotations were measured using Perkin-Elmer 341 polarimeter.

4.2. Microorganisms and the lead compounds

C. blakesleeana NRRL 1369 was obtained from the ARS Culture Collection, USA. *Neosartoria hiratsukae*, an endophytic fungus isolated from *Astragalus angustifolius* (Ekiz, 2016), was also selected for preparative scale biotransformation. Ruscogenins (NR to R ratio was 75:25) and neoruscogenin (>95%) was donated by Bionorm Natural Products Ltd. (Izmir, Turkey).

4.3. Microbial transformation procedures

The biotransformation process was conducted at two scales analytical and preparative. Stock cultures of *C. blakesleeana* and *N. hiratsukae* stored at +4 °C were revived on PDA (Potato-dextrose-agar) slants at 30 °C and 25 °C, respectively. Inoculum of 2% derived from the suspension of 5–7 days old cultures with Tween 80 (0.1%), was used in the biotransformation process. One-stage fermentation protocol was followed where ruscogenins/neoruscogenin were fed to the biotransformation media 72 h after the inoculation. The biotransformation media contained 2% glucose, 0.5% yeast extract, 0.5% NaCl, 0.5% K₂HPO₄ and 0.5% (w/v) peptone (pH 6.0). Analytical scale was conducted using 250 ml flasks containing 50 ml of media and 10 mg of ruscogenins. Every 2 days, samples with a volume of 0.5 ml were taken and centrifuged. Then the supernatants were extracted with ethyl acetate. In order

to differentiate the transformed products from the metabolites of the microorganisms, one flask was kept in same condition with no lead compound. In preparative scale, 1 L flasks containing 200 ml of media and 40 mg of substrate used in same conditions. In the *C. blakesleeana* NRRL 1369 study at preparative scale, 640 mg of ruscogenins was used (30 °C, 180 rpm, 10 days), whereas in the *N. hiratsukae* preparative experiment, 1.48 g of neoruscogenin was fed (25 °C, 180 rpm, 10 days).

4.4. Extraction and isolation

After the incubation period, separation of the microorganisms from the growth media was first step. This process was carried out by using vacuum filtration, and then the broths were extracted with both ethyl acetate (3 times) and *n*-butanol. Ethyl acetate extracts were dried over anhydrous sodium sulfate. Then the organic layers were combined and evaporated under vacuum. By comparing with the reference sample of ruscogenins/neoruscogenin and the blank sample, the biotransformation products were identified on thin layer chromatography (TLC) plates.

Compounds **1**–**5** and **8**, **9** were isolated from the EtOAc extract (712 mg) of *C. blakesleeana* and Ruscogenins. This extract was applied to reverse phase flash column chromatography (40 g, 5 ml/min; 30%; ACN:H₂O), which provided 14 main fractions. Fraction A (27.7 mg) was subjected to normal phase silica gel column (1.5/16 cm; 12 g) using CHCl₃:MeOH:H₂O (95:5:0.5) for the further purifications. This afforded 2 mg **1** and 0.5 mg **2**. To obtain metabolite **8**, fraction B (39 mg) was precipitated in acetonitrile to give 10.5 mg of **8**. The supernatant was evaporated (22 mg), and was applied to open column chromatography containing normal phase silica gel (1.5/14 cm; 10 g) with Hex:EtOAc:MeOH solvent system (10:10:2)

to give 1.5 mg of **3** and 1 mg of **4**. Fraction C (63 mg) was subjected to silica gel (2/13 cm; 15 g) eluting with CHCl₃:MeOH (90:10) to give 15 mg of **9** (mixture of **9a** and **9b**), and 3 mg of **5**. Compounds **6**, **7** and **10** were isolated from the combined EtOAc and BuOH extract (4.87 g) of *N. hiratsukae* and neuroscogenin. The crude extract was divided into two parts, and each was applied to Sephadex-LH 20 column (3/48 cm, 100 g; 100% MeOH). Fractions 31–46 and 30–42, which had NR metabolites, were pooled together for further purification. The combined fraction was subjected to silica gel chromatography (3/27.5 cm, 80 g) eluting with gradient of MeOH in CHCl₃ (increasing polarity from 100:0 to 80:20, CHCl₃:MeOH) and yielded 25 main fractions. The fraction A precipitated by ACN and 20 mg NR have been recovered. For further purification; fraction B (48.5 mg) was applied to VLC containing reverse phase silica gel (3/8 cm; 20 g), eluted with gradient of Acetonitrile:H₂O (%20–50 ACN) to give **6** (11 mg) and **10** (4 mg). The fractions containing metabolite **7** (50 mg) was subjected to silica gel chromatography (2/17 cm, 15 g), and elution with CHCl₃:MeOH (90:10) afforded 15 mg of **7**.

4.5. Compound characterization

7β,14α-dihydroxy neuroscogenin [Spirosta-5,25(27)-dien-1β,3β,7β,14α-tetraol] (**1**): White amorphous solid. $[\alpha]_D^{25}$: –132 (c 0.03, MeOH). ¹H NMR (pyridine-*d*₅, 400 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 100 MHz): see Table 2; HR-MS: *m/z* 461.2899 [M+H]⁺ (C₂₇H₄₁O₆, calcd. 461.2903).

12β,14α-dihydroxy neuroscogenin-7-one [Spirosta-5,25(27)-dien-1β,3β,12β-14-tetraol-7-one] (**2**): White amorphous solid. $[\alpha]_D^{25}$: –91 (c 0.04, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 2; HR-MS: *m/z* 497.2531 [M+Na]⁺ (C₂₇H₃₈O₇Na, calcd. 497.2515).

12β,21-dihydroxy neuroscogenin-7-one [Spirosta-5,25(27)-dien-1β,3β,12β, 21-tetraol-7-one] (**3**): White amorphous solid. $[\alpha]_D^{25}$: –25 (c 0.07, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 2; HR-MS: *m/z* 497.2509 [M+Na]⁺ (C₂₇H₃₈O₇Na, calcd. 497.2515).

7α,12β-dihydroxy ruscogenin [Spirosta-5-en-1β,3β,7α,12β-tetraol] (**4**): White amorphous solid. $[\alpha]_D^{25}$: –122 (c 0.07, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (CD₃OD, 125 MHz): see Table 2; HR-MS: *m/z* 463.3059 [M+H]⁺ (C₂₇H₄₃O₆, calcd. 463.3060).

12β,21-dihydroxy neuroscogenin-1,7-dione [Spirosta-5,25(27)-dien-3β,12β,21-triol-1,7-dione] (**5**) White amorphous solid. $[\alpha]_D^{25}$: –73 (c 0.03, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 2; HR-MS: *m/z* 495.2363 [M+Na]⁺ (C₂₇H₃₆O₇Na, calcd. 495.2359).

12α-hydroxy neuroscogenin [Spirosta-5,25(27)-dien-1β,3β,12α-triol] (**6**) White amorphous solid. $[\alpha]_D^{25}$: –46 (c 0.07, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 2; HR-MS: *m/z* 445.2950 [M+H]⁺ (C₂₇H₄₁O₅, calcd. 445.2954).

Neuroscogenin-1-O-β-glucopyranoside [Spirosta-5,25(27)-dien-1β,3β,diol-1-O-β-glycopyranoside (**7**): White amorphous solid. $[\alpha]_D^{25}$: –33 (c 0.07, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 1; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 2; HR-MS: *m/z* 613.3353 [M+Na]⁺ (C₃₃H₅₀O₉Na, calcd. 613.3353).

7β,12β-dihydroxy neuroscogenin [Spirosta-5,25(27)-dien-1β,3β,7β,12β-tetraol] (**8**): White amorphous solid. $[\alpha]_D^{25}$: –118 (c 0.07, MeOH). ¹H NMR (CD₃OD, 400 MHz): see Table 3; ¹³C NMR (CD₃OD, 100 MHz): see Table 3; HR-MS: *m/z* 483.2717 [M+Na]⁺ (C₂₇H₄₀O₆Na, calcd. 483.2723).

Mixture of 12β-hydroxy neuroscogenin-7-one and 12β-hydroxy ruscogenin-7-one [Spirosta-5,25(27)-dien-1β,3β,12β-triol-7-one and Spirosta-5-en-1β,3β,12β-tetraol-7-one, respectively] (**9a** and **9b**): White amorphous solid. $[\alpha]_D^{25}$: –91 (c 0.07, MeOH). ¹H NMR

(CD₃OD, 500 MHz): see Table 3; ¹³C NMR (CD₃OD, 125 MHz): see Table 3; HR-MS: *m/z* 459.2732 [M+H]⁺ for **9a** (C₂₇H₃₉O₆, calcd. 459.2747) *m/z* 461.2885 [M+H]⁺ for **9b** (C₂₇H₄₁O₆, calcd. 461.2747).

7β-hydroxy neuroscogenin 12-one [Spirosta-5,25(27)-dien-1β,3β,7β-triol-12-one] (**10**): White amorphous solid. $[\alpha]_D^{25}$: –142 (c 0.07, MeOH). ¹H NMR (pyridine-*d*₅, 500 MHz): see Table 3; ¹³C NMR (pyridine-*d*₅, 125 MHz): see Table 3; HR-MS: *m/z* 481.2566 [M+Na]⁺ (C₂₇H₃₈O₆Na, calcd. 481.2566).

Acknowledgments

This project was supported by Ege University Scientific Research Project 15ECZ011 and partly by TÜBİTAK (Project No: 114Z958). We are very grateful to Bionorm Natural Products for providing ruscogenins, and special thanks to Anzarulhaque Anwarulhaque for running NMR experiments at Prince Sattam bin Abdulaziz University.

Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.phytochem.2018.04.002>.

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