arXiv:2202.04144v1 [cond-mat.mtrl-sci] 8 Feb 2022

Electronic, Magnetic and Vibrational Properties of Single Layer Aluminum Oxide

A. Kutay Ozyurt,¹ Deniz Molavali,¹ and Hasan Sahin²

¹Department of Physics, Izmir Institute of Technology, 35430, Izmir, Turkey

²Department of Photonics, Izmir Institute of Technology, 35430, Izmir, Turkey

The structural, magnetic, vibrational and electronic properties of single layer aluminum oxide (AlO_2) are investigated by performing state-of-the-art first-principles calculations. Total energy optimization and phonon calculations reveal that aluminum oxide forms a distorted octahedral structure $(1T'-AlO_2)$ in its single layer limit. It is also shown that surfaces of $1T'-AlO_2$ display magnetic behavior originating from the O atoms. While the ferromagnetic (FM) state is the most favorable magnetic order for $1T'-AlO_2$, transformation to a dynamically stable antiferromagnetic (AFM) state upon a slight distortion in the crystal structure is also possible. It is also shown that Raman activities ($350-400 \text{ cm}^{-1}$) obtained from the vibrational spectrum can be utilized to distinguish the possible magnetic phases of the crystal structure. Electronically, both FM and the AFM phases are semiconductors with an indirect band gap and they can form a type-III vdW heterojunction with graphene-like ultra-thin materials. Moreover, it is predicted that presence of oxygen defects that inevitably occur during synthesis and production do not alter the magnetic state, even at high vacancy density. Apparently, ultra-thin $1T'-AlO_2$ with its stable crystal structure, semiconducting nature and robust magnetic state is a quite promising material for nanoscale device applications.

I. INTRODUCTION

Compounds formed by the combination of aluminum and oxygen atoms have found their use in many areas of technology. Among the possible compounds, Alumina (Al_2O_3) with its high melting point,[1] self-healing crystal structure,[2] high dielectric constant[3] and insulating properties[4, 5] have been widely used in electronic device applications. While bulk Al_2O_3 is a suitable material for high-throughput anti-reflective polymer lenses[6], its thin forms have been shown to improve the spin-polarized electron tunneling in magnetic tunnel junctions.[7] In addition, successful synthesis of its nanowire,[8] quantum dot[9] and thin two-dimensional[10, 11] forms have also been reported so far.

Aluminum oxides can also be stabilized in the form of aluminates (Al_xO_y) . It was demonstrated that thin layers of Lanthanum-based aluminates can serve as a source of two-dimensional electron liquid exhibiting electric-field-tunable superconductivity.[12] It was also shown that using ultrathin Lithium aluminate nanoflakes as a polar additive, having an enormous active surface area to immobilize the lithium polysulfides, leads to extremely stable cycling performance in batteries.[13] Furthermore, one-dimensional forms of strontium-europium-based aluminates are luminescent nanoribbons that can function as both light generators and waveguides, and therefore they are suitable materials for miniaturized photonic circuitry.[14] In addition to experimental studies, stabilization of single layer α -AlO displaying indirect semiconducting behavior was also predicted theoretically.[15]

At the point where nanotechnology has reached, it is necessary to further miniaturize materials, and nowadays, there is also a great need for atomically thin building blocks of all kinds. Following the first successful synthesis of graphene, [16] its unusual properties such as high thermal conductivity, [17] semi-metallic behavior, [18] and ultra-high mechanical strength [19] have been demonstrated in a short period of time. Studies in the following years have revealed that ultra-thin stable phases of other graphene-like materials may also exist. Among them especially h-BN with its insulating properties, [20] MoS₂ with its thickness-dependent electronic and optical properties, [21, 22] ReS₂ for its in-plane anisotropic and thickness-independent properties [23] and silicene with its ultra-high carrier mobility [24] have been the most prominent ones.

Although many graphene-like materials have been successfully synthesized to date, very few of them show magnetic properties. Recently, it has been found that the itinerant ferromagnetism persists in 2D Fe₃GeTe₂ down to the monolayer with an out-of-plane magnetocrystalline anisotropy. [25] In addition, intrinsic ferromagnetic half-metallic behavior of monolayer square $CrBr_2$ and the strain-dependent phase transitions were predicted using first-principles calculations. [26] Very recently, growth of iron chloride (FeCl₂) films on Au(111) and graphite with atomic thickness by molecular-beam epitaxy (MBE) technique has also been reported. [27] Besides, we showed that the magnetic ground state of single layer FeCl₂ crystal can be modulated by the point defects and structural stability was maintained despite the heavy defect doping. [28]

In this work, we investigated the ultra-thin graphene-like crystal structure of aluminum oxide (AlO_2) by means of abinitio calculations. The rest of the paper is organized as follows; computational methodology is given in Sec. II. Possible magnetic phases and their vibrational characteristics are investigated in Sec. III A. In addition, electronic properties of AFM and FM phases and their possible vdW type heterostructures are investigated in Sec.III B. Formation of

TABLE I. For the different magnetic phases of 1T'-AlO₂: lattice constants for rectangular unit cell (a and b), thickness (h), the amount of donated (received) electron $(\rho_{A1} \text{ and } \rho_O)$, the work function (ϕ) , calculated cohesive energy for per atom (E_{Coh}) , the band gap value (E_{Gap}) , total magnetic moment for per the formula, (μ) , in-plane stiffness along zigzag direction, (C_{zz}) , in-plane stiffness along armchair direction, (C_{arm}) , Poisson ratio along zigzag direction, (v_{zz}) , Poisson ratio along armchair direction, (v_{arm}) , calculated hole effective mass (m_h^*) , calculated electron effective mass (m_e^*) where m₀ is the free electron mass.

| | a (Å) | b (Å) | h (Å) | ρ_{A1} (e ⁻) | ρ _Ο (e ⁻) | ϕ (eV) | E_{Coh} (eV/atom) | $\begin{array}{c} \mathbf{E}_{Gap} \\ (\mathrm{eV}) \end{array}$ | $\mu \ (\mu_B)$ | C_{zz} (N/m) | C_{arm} (N/m) | v_{zz} | Varm - | ${f m}_h^* \ ({f m}_0)$ | ${f m}_e^* \ ({f m}_0)$ |
|--|----------------|--------------|----------------|---|-------------------------------------|--------------|---------------------|--|---------------------------------------|---|---|----------------|---|-------------------------|-------------------------|
| $\begin{array}{c} \mathrm{FM} \\ \mathrm{AFM} \end{array}$ | $3.05 \\ 3.06$ | 4.84 4.82 | $1.77 \\ 1.78$ | $\begin{array}{c} 0.54 \\ 0.54 \end{array}$ | 7.23 7.23 | 8.01 7.96 | $5.57 \\ 5.56$ | $\begin{array}{c} 0.63 \\ 1.18 \end{array}$ | $\begin{array}{c} 1 \\ 0 \end{array}$ | $\begin{array}{c} 126 \\ 108 \end{array}$ | $\begin{array}{c} 138 \\ 123 \end{array}$ | $0.40 \\ 0.33$ | $\begin{array}{c} 0.36 \\ 0.37 \end{array}$ | 2.29 1.20/18.34 | $1.33 \\ 8.52$ |



FIG. 1. (a) Energy versus lattice constant (a) of 2D 1T'-AlO₂. Top views of the geometric structures of (b) FM and (c) AFM state where blue and red atoms stand for Al and O, respectively. For 3D plot of magnetization charge densities minority and majority states are shown by blue and yellow color.

single O vacancy and resulting modifications in electronic properties are given in Sec.III C. Finally, the conclusions are presented in Sec.IV.

II. COMPUTATIONAL METHODOLOGY

For determination of optimized crystal structures and resulting electronic and magnetic characteristics of AlO₂ firstprinciples calculations were performed by using Vienna ab initio Simulation Package (VASP).[29–31] The exchangecorrelation potential was treated using the Perdew-Burke-Ernzerhof Generalized Gradient Approximation (PBE-GGA).[32] The projector augmented wave (PAW) potentials were used as pseudo-potentials.[33] Van der Waals (vdW) correction within the form of DFT-D2 method of Grimme is used for approximation of dispersion forces.[29] Bader charge technique was used for determination of charge for each atom in the crystal structure.[34]

The kinetic energy cutoff and the convergence criteria for the total energy were set to 500 eV and 10^{-5} eV, respectively. At least 18.2 Å vacuum spacing was inserted in order to avoid adjacent layer-layer interactions along the vertical direction. For the structural optimization of the primitive unit cell, k-space was sampled by using $9 \times 9 \times 1$



FIG. 2. Phonon band structure of $1T'-AlO_2$ calculated for (a) the FM and (b) the AFM states. (c) The Raman activity spectrum of the zone-centered phonon modes. (d) The representative images of Raman active phonon vibrations.

k-point mesh, and doubled to $18 \times 18 \times 1$ in density of states calculations. Structural optimizations were carried on until the pressure is in between ± 1 kB along the each lattice direction. The cohesive energy per atom was calculated using the equation $E_{Coh} = \frac{1}{n_{tot}} [n_{Al}E_{Al} + n_O E_O - E_{SL}]$ where n_{Al} (n_O) is the number of Al (O) atoms in the system, while E_{Al} and E_O refer to individual energies of Al and O atoms, respectively. E_{SL} is the total energy of the corresponding single layer and n_{tot} is the total number of atoms per unit cell.

The vibrational properties were investigated via PHONOPY code, which uses the small displacement method to generate the force constant matrix.[35] In addition, the zone-centered phonon frequencies and the corresponding off-resonant Raman activities were calculated using the finite-difference method. Basically, in a Raman experiment instantly scattered photons, whose dispersion with respect to a shift in frequency gives the Raman spectrum, are collected. The treatment of Raman activities of each vibration is based on Placzek's classical theory of polarizability in which the activity of a Raman active mode is proportional to the square of the term, $|\hat{e}_s R \hat{e}_i|$ where \hat{e}_s and \hat{e}_i stand for the polarization vectors of the scattered and the incident light, respectively. In addition, R represents the Raman tensor whose elements are the derivatives of the polarizability of the material with respect to the phonon normal modes. Notably, for a Raman active phonon mode some of the elements of Raman tensor are non-zero.

III. RESULTS

A. Structural, Magnetic and Vibrational Properties

Characteristic vibrational, electronic and magnetic fingerprints of any material are directly determined by its crystal symmetry. Therefore, a reliable prediction of the ground state crystal structure is of importance. Total energy optimization calculations show that (see Fig.1(a)) the AlO₂ crystal has a six-atom rectangular unit cell with the lattice parameter of **a** (**b**) is 3.05 (4.84) Å. Here the optimized structure of AlO₂ displays slightly distorted 1T hexagonal lattice (namely 1T'-AlO₂), where the central Al subplane is surrounded by an octahedral type arrangement of O atoms. It worths mentioning that 1T'-AlO₂ displays ferromagnetic (FM) ground state with 1 μ_B per formula. Magnetic charge density distribution (ρ_{\uparrow} - ρ_{\downarrow}) shown in Fig. 1(b) reveals that the ferromagnetic behavior stems from p_z orbitals of O atoms. However, there is also a possible antiferromagnetic (AFM) phase of the crystal structure which is ~40 meV less preferable than the FM one. Here cohesive energy of FM and AFM phase is calculated to be 5.57 and 5.56 eV/atom, respectively. In the AFM phase, that emerges as a result of the neighboring oxygens approaching each other, ferromagnetically ordered AlO₂ chains are formed and adjacent chains prefer to be antiferromagnetically ordered with each other. Compared to the FM phase, as a result of closer packing of AlO₂ chains in the AFM phase, there is a narrowing in the **b** lattice parameter and a slight increase in the layer thickness (defined as the distance between the neighboring subplanes of O atoms).

Bader charge analysis shows that crystal lattice is constructed with Al-O bonds having ionic character, where the charge donation from Al to each O is $1.23 e^{-1}$ for both FM and AFM states. Additionally, the work functions of the FM

and AFM phases are calculated to be 8.01 and 7.96 eV, respectively, which are significantly higher than that of single layer structures of MoS_2 (6.11 eV), WS_2 (5.89 eV), $MoSe_2$ (5.49 eV) and h-BN (4.7 eV).[36, 37] Apparently, such materials having high work function are desirable for their use as charge extraction and injection layers in electronic and optical devices in order to improve device capability.[38, 39]

For a discussion on resistance of the 2D crystal structure distortion under mechanical load, determination of in-plane stiffness (C) and the Poisson ratio (v) is of importance. Due to its anisotropic nature, the 1T' phase of AlO₂ possesses anisotropy by means of its elastic constants. Here, the two elastic constants along with the zigzag (zz) and armchair (arm) directions are calculated by using the equations [40] $C_{zz} = \frac{1}{C_{22}} [C_{11}C_{22} - C_{12}^2]$ and $C_{arm} = \frac{1}{C_{11}} [C_{11}C_{22} - C_{12}^2]$. Accordingly, in-plane stiffness values of FM and AFM phases are found to be 126(138) and 108(123) N/m for FM and AFM phases along the zz(arm) directions, respectively. Compared with the single layer Al₂O₃ (48 N/m)[11], AlO₂ seems to be a much stiffer material. On the other hand, as compared to single layer of AlN(116 N/m)[41], AlO₂ phases are found to possess similar mechanical features in terms of the in-plane stiffness.

The Poisson ratio, which is defined as the ratio of the transverse contraction strain to the longitudinal extension, describes the mechanical response of a material to the external load. The Poisson ratio through the zz and arm directions are calculated by using the equations[40] $v_{zz} = \frac{C_{12}}{C_{22}}$ and $v_{arm} = \frac{C_{12}}{C_{11}}$. It is known that for a stable linear elastic material, the Poisson ratio varies between -1.0 and 0.5.[42] Here, the Poisson ratios of FM (AFM) phases of 1T'-AlO₂ are calculated to be 0.40 (0.33) along zz direction and 0.36 (0.37) along arm direction. These values indicate the soft nature of AlO₂ when compared with those of the well-known 2D structures, such as graphene and MoS₂, for which the Poisson ratios are reported to be 0.19 and 0.26, respectively.[19, 43]

The dynamical stability of 1T'-AlO₂ in FM and AFM states is also examined through their phonon band dispersions, which are obtained by utilizing the small displacement method. Phonon dispersions depicted in Fig.2 show that all the phonon branches have real eigenfrequencies through the whole Brillouin zone and therefore both FM and AFM phases of 1T'-AlO₂ crystal are dynamically stable. It is seen that distorted crystal structure of 6-atom primitive unit cell of the crystal structure leads to presence of 18 phonon branches. The highest frequency optical phonon branch is found at 622 (625) cm⁻¹ for the FM (AFM) order which is lower than those single layer AlN (~ 900cm⁻¹), Al₂O₃ (1500cm⁻¹), and α -AlO (~ 650cm⁻¹).[11, 15, 41]

For further discussion of vibrational characteristics, zone-centered first-order off-resonant Raman spectra of FM and AFM phases are also calculated. As shown in Fig. 2(c-d), The Raman active modes labeled by numbers I-II, which have the frequencies of 374.8 (401) cm⁻¹ for FM (AFM) states, are attributed to the shear-like vibration of upper and lower O atoms towards the Al atoms. In addition, the phonon modes III-IV are calculated to be at the frequencies 477 (482) cm⁻¹ for FM (AFM) states and are shown to arise from the shear-like vibration of O atoms parallel to the Al atoms. Moreover, the Raman active modes, V-VI having frequencies 582 (583) cm⁻¹ for FM (AFM), represent the out-of-plane vibration of upper and lower O atoms with respect to each other. While the two magnetic phases are distinguishable in terms of the frequencies of the Raman active modes, the calculated activities reveal their distinct features. The Raman activities of the most prominent modes (I-II) are found to be the same while the activities of III and V in the FM phase are found to be lower as compared to that of IV and VI in the AFM state.

B. Electronic Properties

Electronic characteristics of FM and AFM phases of single layer 1T'-AlO₂ are investigated through their electronic band dispersions and the band decomposed charge densities. As shown in Fig. 3(a), FM phase of AlO₂ is a ferromagnetic semiconductor with an indirect bandgap of 0.63 eV. As a consequence of the large splitting in majority and minority states valence band maximum (VBM) and conduction band minimum (CBM) of the FM phase is composed of the same spin type. While the edge of the VBM reside at the Y- Γ point, the CBM is located at the Brillouin zone center. It is also seen that both VBM and CBM band edge states have the same tilted O- p_z character. As shown in Table I, the hole effective mass at the vicinity of VBM is found to be 2.29 m₀, while the electron effective mass at the CBM is calculated to be 1.33 m₀ (where m₀ is electron mass).

In addition, AFM phase displays indirect semiconducting behavior with 1.18 eV bandgap where the VBM and CBM are located at the X and the Γ of the BZ, respectively. It appears that VBM states stem from orbitals perpendicularly oriented to the crystal surface leads to the formation of quite flat bands around the X point and therefore relatively large hole effective mass for the AFM state. Moreover, the hole effective mass is found to be anisotropic and two different values are calculated along with the Γ -X and X-S directions (1.20 and 18.34 m₀, respectively) while the electron effective mass at the vicinity of the CBM is found to be 8.52 m₀. Apparently, lattice contraction originating from the formation of AlO₂ chains leads to the much heavier charge carriers in the AFM phase.

Considering a nano device in which a 2D ultra-thin material is used as the building block, it is clear that within the device architecture the material will form interfaces with other materials that can be thought as heterostructures. Here, we also investigate the vertical vdW heterostructures of 1T'-AlO₂ phases constructed with the 2D crystals such



FIG. 3. Electronic band structure of $1T'-AlO_2$ for (a) FM and (b) AFM phase. Dashed and solid lines show the majority and minority spin states which are degenerate in AFM state. (c) Comparative band alignments of $1T'-AlO_2$ phases with MoS₂, WSe₂, InO and AlN single layers.

as MoS_2 , WSe_2 , InO, and AlN. As presented in Fig. 3(c), as a consequence of their high photoelectric threshold, energies of VBM and the CBM of both phases of $1T'AlO_2$ are much lower than that of the other materials considered. Therefore, it can deduced that either FM- or AFM-phase of $1T'-AlO_2$ form type-III heterojunction with single layers of MoS_2 , WSe_2 , InO, and AlN. Obviously, such type-III vdW heterostructures might exhibit a negative differential resistance phenomenon due to the existence of band-to-band tunneling as the electrons have a tendency to tunnel from VBM of these crystal structures to the CBM of $1T'AlO_2$.[44–46]

FIG. 4. The atomic structure, magnetization charge density and corresponding electronic density of states of single oxygen defected 1T'-AlO₂ in (a-b) FM and (c-d) AFM phase. The missing atom is shown by dashed red circle.

C. Effect of Oxygen Vacancies

Whether the synthesis technique is a top-down or bottom-up approach, the formation of point defects in twodimensional crystals is inevitable. In order to investigate the effect of atomic vacancies on the electronic properties of single layer 1T'-AlO₂, the formation of a single O vacancy, V_O, is considered and the defected crystal structure is fully optimized. In order to simulate the single O vacancy, a supercell structure containing 96 atoms is considered and an O atom is removed from the layer. After full optimization of the O-vacant single layer of AlO₂, it is shown that structural deformations around the defect site are compensated by the local distortion of three Al atoms which tend to form Al-dimers as the hole gets larger via O vacancy. While the Al-Al distance is 2.84 Å in the host lattice, that of between the Al atoms around the defect site is 2.75 Å as a result of the structural distortion. On the other hand, by the formation of the vacancy, O atoms around the defect site move towards the Al atoms surrounding the defect site and thus, the Al-O bond lengths get smaller. It is also found that the formation of V_O shrinks the lattice by 0.4% along the x-direction while it locally enlarges the host lattice by 0.3%.

The defect formation energy is calculated using the relation; $E_f = [E_{tot}(def) + E_O] - E_{tot}(pristine)$ where $E_{tot}(def)$ and E_{tot} represents the total energies of the defected and the pristine single layers, respectively while E_O stands for the single O energy. Here, the calculated defect formation energy of V_O is 6.24 eV which indicates that it is very difficult to subsequently create an O vacancy in the single layer 1T'-AlO₂.

Moreover, magnetic ground state calculations reveal that FM interaction is still energetically more favorable in defected $1T'-AlO_2$. It is found that the formation of O vacancy leads to a decrease in the total magnetic moment of the structure $(2\mu_B)$. Apparently, the creation of a single O vacancy induces local magnetic moments which are altered by the deviations from the nominal stoichiometry. The effect of single O vacancy on the electronic properties

of 1T'-AlO₂ is investigated in terms of the total electronic density of states. Apparently, O vacancy induces midgap states that result in a transition to the metallic state in the oxygen vacant domains of the 1T'-AlO₂ crystal. As shown in Fig. 4(c), similar to the FM phase, the presence of the oxygen vacancy does not cause a remarkable change in the magnetic ground state when the crystal structure is in the AFM phase. It is also seen that while the magnetic order of the AFM remains unchanged, a magnetic moment of $2\mu_B$ per oxygen occurs because the balance between the sublattice atoms is disturbed due to the missing oxygen atom. Additionally, as shown Fig. 4(d) formation of oxygen vacancies also leads to a semiconductor-to-metal transition at the vicinity of defect domains.

IV. CONCLUSIONS

In conclusion, the first-principles calculations were performed on single layer AlO_2 and its structural, magnetic, vibrational and the electronic properties were investigated. Total energy optimizations revealed that single layer 1T'- AlO_2 possesses two distinct magnetic phases, namely FM and AFM. While the FM phase of 1T'- AlO_2 was found to be energetically more favorable over the AFM one, both phases are dynamically stable and they are distinguishable in terms of their Raman spectra. Electronically, both FM- and AFM-1T'- AlO_2 structures were shown to be semiconductors with an indirect band gap. It is also seen that 1T'- AlO_2 crystals form a broken-gap type-III heterojunction when combined with other graphene-like ultra-thin materials. Moreover, we predicted that presence of oxygen vacancies yields semiconductor-to-metal transition without altering the magnetic order in the crystal structure. The ultra-thin 1T'- AlO_2 with its semiconducting nature and robust magnetic ground state unaffected by the presence of structural defects is a promising material for nanoscale device applications.

V. ACKNOWLEDGMENTS

Computational resources were provided by TUBITAK ULAKBIM, High Performance and Grid Computing Center (TR-Grid e-Infrastructure).

- Mavrič, A.; Fanetti, M.; Mali, G.; Valant, M. High-temperature stabilization of bulk amorphous Al₂O₃. J. Non-Cryst. Solids 2018, 499, 363–370.
- [2] Yang, Y.; Kushima, A.; Han, W.; Xin, H.; Li, J. Liquid-like, self-healing aluminum oxide during deformation at room temperature. Nano Lett. 2018, 18, 2492–2497.
- [3] Peacock, P.; Robertson, J. Band offsets and Schottky barrier heights of high dielectric constant oxides. J. Appl. Phys. 2002, 92, 4712–4721.
- [4] Vishvakarma, S.; Kannan, R.; Vashishtha, N.; Pal, A. S. Evaluation of hybrid sol-gel alumina coating for electrical resistance applications. Surf. Topogr. 2021, 9, 045030.
- [5] Kim, J.; Ha, J.-H.; Lee, J.; Song, I.-H. Optimization for permeability and electrical resistance of porous alumina-based ceramics. J. Korean Ceram. Soc. 2016, 53, 548–556.
- [6] Yanagishita, T.; Nishio, K.; Masuda, H. Anti-reflection structures on lenses by nanoimprinting using ordered anodic porous alumina. Appl. Phys. Express 2009, 2, 022001.
- [7] Wilt, J.; Goul, R.; Acharya, J.; Sakidja, R.; Wu, J. Z. In situ atomic layer deposition and electron tunneling characterization of monolayer Al₂O₃ on Fe for magnetic tunnel junctions. AIP Adv. 2018, 8, 125218.
- [8] Meng, F. C.; Liu, C.; Zhang, X. L.; Ren, H.; Jia, T. Synthesis and Characterization of γ-Al₂O₃ Nanowires by a Surfactant Assisted Precipitation Method. *Key Eng. Mater.* 2014, 602-603, 93 – 96.
- [9] Li, Z.; Kong, L.; Huang, S.; Li, L. Highly luminescent and ultrastable CsPbBr₃ perovskite quantum dots incorporated into a silica/alumina monolith. Angew. Chem 2017, 129, 8246–8250.
- [10] Napari, M.; Huq, T.; Meeth, D.; Heikkilä, M.; Niang, K.; Wang, H.; Iivonen, T.; Wang, H.; Leskelä, M.; Ritala, M.; Flewitt, A.; Hoye, R.; MacManus-Driscoll, J. Role of ALD Al 2 O 3 Surface Passivation on the Performance of p-Type Cu₂O Thin Film Transistors. ACS Appl. Mater. Interfaces 2021, 13, 4156–4164.
- [11] Song, T.; Yang, M.; Chai, J.; Callsen, M.; Zhou, J.; Yang, T.; Zhang, Z.; PAN, J.; Chi, D.; Feng, Y. P.; Wang, S. The stability of aluminium oxide monolayer and its interface with two-dimensional materials. *Sci. Rep.* 2016, 6, 29221.
- [12] Cheng, G.; Tomczyk, M.; Lu, S.; Veazey, J.; Huang, M.; Irvin, P.; Ryu, S.; Lee, H.; Eom, C.-B.; Hellberg, C.; Levy, J. Electron pairing without superconductivity. *Nature (London)* 2015, 521, 196–9.
- [13] Ghosh, A.; Kumar, A.; Roy, A.; Nguyen, C.; Ahuja, A.; Adil, M.; Chatti, M.; Kar, M.; MacFarlane, D. R.; Mitra, S. Ultrathin Lithium Aluminate Nanoflake-Inlaid Sulfur as a Cathode Material for Lithium–Sulfur Batteries with High Areal Capacity. ACS Appl. Energy Mater. 2020, 3, 5637–5645.

- [14] Li, X.; Budai, J. D.; Liu, F.; Chen, Y.-S.; Howe, J. Y.; Sun, C.; Tischler, J. Z.; Meltzer, R. S.; Pan, Z. Crystal structures and optical properties of new quaternary strontium europium aluminate luminescent nanoribbons. J. Mater. Chem. C 2015, 3, 778–788.
- [15] Demirci, S.; Avazh, N.; Durgun, E.; Cahangirov, S. Structural and electronic properties of monolayer group III monochalcogenides. Phys. Rev. B 2017, 95, 115409.
- [16] Novoselov, K. S.; Geim, A. K.; Morozov, S. V.; Jiang, D.-e.; Zhang, Y.; Dubonos, S. V.; Grigorieva, I. V.; Firsov, A. A. Electric field effect in atomically thin carbon films. *Science* 2004, 306, 666–669.
- [17] Geim, A. K. Graphene: status and prospects. Science 2009, 324, 1530–1534.
- [18] Neto, A. C.; Guinea, F.; Peres, N. M.; Novoselov, K. S.; Geim, A. K. The electronic properties of graphene. Rev. Mod. Phys 2009, 81, 109.
- [19] Lee, C.; Wei, X.; Kysar, J. W.; Hone, J. Measurement of the elastic properties and intrinsic strength of monolayer graphene. Science 2008, 321, 385–388.
- [20] Li, L. H.; Chen, Y. Atomically thin boron nitride: unique properties and applications. Adv. Funct. Mater. 2016, 26, 2594–2608.
- [21] Mak, K. F.; Lee, C.; Hone, J.; Shan, J.; Heinz, T. F. Atomically thin MoS₂: a new direct-gap semiconductor. *Phys. Rev. Lett.* 2010, 105, 136805.
- [22] Wang, Q. H.; Kalantar-Zadeh, K.; Kis, A.; Coleman, J. N.; Strano, M. S. Electronics and optoelectronics of two-dimensional transition metal dichalcogenides. *Nat. Nanotechnol.* 2012, 7, 699–712.
- [23] others, et al. Monolayer behaviour in bulk ReS_2 due to electronic and vibrational decoupling. Nat. Commun. 2014, 5, 1–6.
- [24] Cahangirov, S.; Topsakal, M.; Aktürk, E.; Şahin, H.; Ciraci, S. Two-and one-dimensional honeycomb structures of silicon and germanium. Phys. Rev. Lett. 2009, 102, 236804.
- [25] Deng, Y.; Yu, Y.; Song, Y.; Zhang, J.; Wang, N. Z.; Sun, Z.; Yi, Y.; Wu, Y. Z.; Wu, S.; Zhu, J.; et al., Gate-tunable room-temperature ferromagnetism in two-dimensional Fe₃GeTe₂. *Nature (London)* **2018**, 563, 94–99.
- [26] Li, F.; Ren, Y.; Wan, W.; Liu, Y.; Ge, Y. Strain tunable intrinsic ferromagnetic in 2D square CrBr₂. AIP Adv. 2021, 11, 115220.
- [27] Xuhan, Z. et al. Atomically Thin 1T-FeCl₂ Grown by Molecular-Beam Epitaxy. J. Phys. Chem. C 2020, 124, 9416–9423.
- [28] Ceyhan, E.; Yagmurcukardes, M.; Peeters, F.; Sahin, H. Electronic and magnetic properties of single-layer FeCl₂ with defects. *Phys. Rev. B* **2021**, 103, 014106.
- [29] Grimme, S.; Antony, J.; Ehrlich, S.; Krieg, H. A consistent and accurate ab initio parametrization of density functional dispersion correction (DFT-D) for the 94 elements H-Pu. J. Chem. Phys. 2010, 132, 154104.
- [30] Kresse, G.; Furthmüller, J. Efficient iterative schemes for ab initio total-energy calculations using a plane-wave basis set. Phys. Rev. B 1996, 54, 11169.
- [31] Kresse, G.; Hafner, J. Ab initio molecular dynamics for liquid metals. Phys. Rev. B 1993, 47, 558.
- [32] Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized gradient approximation made simple. Phys. Rev. Lett. 1996, 77, 3865.
- [33] Blöchl, P. E. Projector augmented-wave method. Phys. Rev. B 1994, 50, 17953.
- [34] Henkelman, G.; Arnaldsson, A.; Jónsson, H. A fast and robust algorithm for Bader decomposition of charge density. *Comput. Mater. Sci.* 2006, 36, 354–360.
- [35] Togo, A.; Tanaka, I. First principles phonon calculations in materials science. Scr. Mater. 2015, 108, 1–5.
- [36] Lanzillo, N. A.; Simbeck, A. J.; Nayak, S. K. Strain engineering the work function in monolayer metal dichalcogenides. J. Phys. Condens. Matter 2015, 27, 175501.
- [37] Pierucci, D.; Zribi, J.; Henck, H.; Chaste, J.; Silly, M.; Bertran, F.; Fevre, P.; Gil, B.; Summerfield, A.; Beton, P.; Novikov, S.; Cassabois, G.; Rault, J.; Ouerghi, A. Van der Waals epitaxy of two-dimensional single-layer h-BN on graphite by molecular beam epitaxy: Electronic properties and band structure. *Appl. Phys. Lett.* **2018**, *112*, 253102.
- [38] Bivour, M.; Temmler, J.; Steinkemper, H.; Hermle, M. Molybdenum and tungsten oxide: High work function wide band gap contact materials for hole selective contacts of silicon solar cells. Sol. Energy Mater. Sol. Cells 2015, 142, 34–41.
- [39] Schulz, P.; Tiepelt, J. O.; Christians, J. A.; Levine, I.; Edri, E.; Sanehira, E. M.; Hodes, G.; Cahen, D.; Kahn, A. High-work-function molybdenum oxide hole extraction contacts in hybrid organic-inorganic perovskite solar cells. ACS Appl. Mater. Interfaces 2016, 8, 31491–31499.
- [40] Ma, Y.; Kou, L.; Huang, B.; Dai, Y.; Heine, T. Two-dimensional ferroelastic topological insulators in single-layer Janus transition metal dichalcogenides M SSe (M= Mo, W). Phys. Rev. B 2018, 98, 085420.
- [41] Şahin, H.; Cahangirov, S.; Topsakal, M.; Bekaroglu, E.; Akturk, E.; Senger, R. T.; Ciraci, S. Monolayer honeycomb structures of group-IV elements and III-V binary compounds: First-principles calculations. *Phys. Rev. B* 2009, *80*, 155453.
- [42] Gercek, H. Poisson's ratio values for rocks. Int. J. Rock Mech. Min. Sci. 2007, 44, 1–13.
- [43] Yagmurcukardes, M.; Senger, R. T.; Peeters, F. M.; Sahin, H. Mechanical properties of monolayer GaS and GaSe crystals. *Phys. Rev. B* 2016, 94, 245407.
- [44] Ionescu, A. M.; Riel, H. Tunnel field-effect transistors as energy-efficient electronic switches. Nature (London) 2011, 479, 329–337.
- [45] Lu, H.; Seabaugh, A. Tunnel field-effect transistors: State-of-the-art. IEEE J. Electron Devices Soc. 2014, 2, 44-49.
- [46] Xia, C.; Du, J.; Li, M.; Li, X.; Zhao, X.; Wang, T.; Li, J. Effects of Electric Field on the Electronic Structures of Broken-gap Phosphorene/SnX₂ (X= S, Se) van der Waals Heterojunctions. *Phys. Rev. Appl.* **2018**, *10*, 054064.